CAPITAL UNIVERSITY OF SCIENCE AND TECHNOLOGY, ISLAMABAD



Determination of Efficacy of *Nigella sativa* Bioactive Compounds Against ALK and EML4 For The Treatment of Lungs Cancer.

by

Rubab Sikandar

A thesis submitted in partial fulfillment for the degree of Master of Science

in the

Faculty of Health and Life Sciences Department of Bioinformatics and Biosciences

2022

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CERTIFICATE OF APPROVAL

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"Trust in the LORD with all your heart and lean not on your understanding; in all your ways acknowledge him and he shall direct your paths". (Proverbs 3:5, 6)

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(Rubab Sikandar)

Abstract

Lungs cancer is the most common cancer type in the world. Due to the developments in diagnostic techniques, therapeutic approaches and traditional remedies, the survival rate has increased but the adverse effects of treatment strategies are also considerable. People around the world are becoming more concerned to use natural products than synthetic drugs. That is the reason for doing this research to explore the potential anticancer agents from Nigella sativa. Fifteen bioactive compounds namely Isoquinoline, β -pinene, Apigenin, Salfredin B11, Pyrazole, Pyragallol, Salicylic acid, Syringic acid, Gallic acid, Camphene, 3,4-dihydroxybenzoic acid, 4-dihydroxycinnamic acid, Caffeic acid, Myristic acid and Stearic acid were selected as ligands. All these ligands were screened out on the basis of Lipinski's rule of five and by studying their ADME properties. Virtual screening of the above mentioned ligands were carried out against Anaplastic Lymphoma kinase and Echinoderm Microtubule associated Protein Like-4 by online tool CB-Dock. Crizotinib and Paclitaxel were selected as standard drugs for comparison. Lead compounds were selected from the above mentioned ligands, which were less toxic than the selected drugs. Visualization analysis studies related to the interaction of selected compound and the drugs were performed by using PyMol and LIGPLOT+ tools. After the complete analysis, Apigenin and Salfredin B11 were selected as potential anticancer compounds which can be considered in future as drug candidate for the treatment of lungs cancer. However further research is required to elucidate their potential medical use.

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Abbreviations

ADMET: Adsorption Distribution Metabolism Excretion **ALK:** Anaplastic Lymphoma Kinase **BBB:** Blood Brain Barrier BCL2: B-cell lymphoma 2 **CB-Dock:** Cavity-Detection guided Blind Docking **CNS:** Central Nervous System **CT:** Computed Tomography CYP2D6: Cytochrome P450 2D6 **EFGR:** Epidermal growth factor receptor EML4: Echinoderm Microtubule Associated Protein Like 4 **EMN:** Electromagnetic navigation FDA: Food Drug Authority GPCRs: G-protein coupled receptors **GRAVY:** Grand average of hydropathicity **HBA:** Hydrogen bond acceptor **HBD:** Hydrogen bond donor **hERG:** Human Ether-a-go-go-Related Gene HIV: Human immunodeficiency virus **II:** Instability Index **KRAS:** Kristin rat sarcoma MAPK: Mitogen-activated protein kinase **MRI:** Magnetic resonance imaging MRTD: Maximum rate tolerated dose **MW:** Molecular weight

NR: Total number of negative charged residues *N. sativa*: *Nigella sativa*NSCLC: Non-small cell lung cancer
OCT: Organic cation transporter
PDB: Protein Data Bank
PR: Total number of positive charged residues
pI: Theoretical pI
R-EBUS: Radial endobronchial ultrasound
SCLC: Small cell lung cancer
TTNA: Transthoracic needle aspiration
VDss: Volume distribution at steady state

Chapter 1

Introduction

1.1 Background

Cancer is the most leading and prominent cause of death at these days. The mortality ratio increases more than 10 million deaths per year [1]. This dramatically increase in cancer incidence rate may be due to many factors like eating habits, age, lifestyle , changing reproductive trends, obesity or overweightness, tobacco use [2]. Many countries of world including developed and developing countries, different types of cancer effect males and females like breast cancer is common in females, lung cancer is common in males. More types of cancer like colorectal cancer, liver, stomach, cervical cancer and prostate cancer are also the cause of death in many developed and under-developed countries [3].

In developed countries Tobacco is the main leading factor that is responsible for 85% cases of lung cancer. Recent statistics shows that over 1.80 million deaths may occur due to lung cancer by 2020 [4].

Lung cancer is the chief cause of death, approximately making 25% death rate due to cancer. Lung cancer is distinguished by uncontrolled growth of cells. This proliferation of cells produces abnormal cellular mass. This mass or tumor grows to different other regions and, with time, attains metastatic capacity and spreads to other parts, at the end cause death. The cancer cell arises due to mutation in genetic complement or may be due to environmental factors [5]. Lung cancer is the common malignant type of cancer around the globe. Oncogenic fusion genes EML4 and ALK are present in Non-Small Lung Cancer cells (NSCLC), which represent almost 7% of such kind of cancers. In cancer disease, ALK proteins have an essential role to withstand the process of apoptosis.

Lungs cancer is the severe form of Cancer that ratio of spread increase day after day. Oncogenic fusion genes are basically the EML4 and ALK kinases that are found in non-small cell lung cancers, making 7% of such tumors. ALK kinases play an important role to halt the apoptotic phenomenon in cancer disease. The fusion of anaplastic lymphoma kinase (ALK) and echinoderm microtubule associated protein-like 4 (EML4) has lately been identified in non-small cell lung cancers (NSCLC) [6].

Many anticancer drugs are used to treat the lung cancer disease. Avastin is the drug approved by FDA because of its improved survival significance when added to several standard first line chemotherapy regimes in non-small cell lung cancer [7]. Erlotinib is also used for treating patients with EGFR mutant non-small lung cancer (NSCLC) [8]. Gefitinib is viable treatment for non-small cell lung cancer (NSCLC) patients whose growths harbor physical transformations in EGFR [9]. Gefitinib and Erlotinib are best in never-smokers with NSCLC having antitumor activity [10]. Interleukin-8 has been widely implicated in processes like angiogenesis and metastasis in lung cancer. In lung cancer, the adenocarcinoma cells produce significant amount of Interleukin-8 [11]. IL-8 increased the growth of non-small cell lung cancer cells, involving activation of epidermal growth factor receptors (EGFR). This receptor plays a key role in proliferation of cells in lung cancer. EFGR receptor and ligand leads to signaling events like mitogen activated protein kinase (MAPK) activation. Transactivation of receptor and EGFR ligand occur due to many G-protein coupled receptors (GPCRs) and involves metalloproteinasemediated membrane bound EGFR ligands [12]. The activity of IL-8 to take part in cell proliferation is blocked by EGFR tyrosine kinase. This is done by specific antibody called anti-EGFR antibody and a metalloproteinase inhibitor. Resistance of MAPK also blocks the effect of IL-8 [13]. With the advancement in science and technology, new techniques are invented that help reduce the rate to cancer and increase the survival of cancer patients. Although with the availability of suitable drug for the disease to prevent it but there are some side effects also that the patients develop resistance against the pharmaceutical drugs. So for better treatment process, there is a need to use medicinal drugs obtained from medicinal plants.

Nigella genus has been an important component of traditional medicine like Unani and Tibb. Because of its marvelous healing power, it has been ranked at the top among medicinal plants [14]. Nigella species are broadly used as a medicinal plant for their therapeutic properties. Seeds and the oils extracted from seeds have old history of use in different frameworks of medications and also utilized as eatables [15]. Extracts from *Nigella* species have mitigating action, principally due to the presence of numerous bioactive molecules. Natural oil obtained from seeds act as secondary metabolites and these metabolites are some sort of monoterpenes [16]. Important therapeutic effects of *Nigella* species include anti-cancer, antioxidant, hypotensive, hepatoprotective, spasmolytic, anti-inflammatory, and bronchodilator, hypoglycemic, hypolipidemic, anti-allergic and immunomodulating properties [17]. Alkaloids, steroids, carbohydrates, flavonoids, fatty acids, etc. are the secondary metabolites found in Nigella genus [18]. Molecular Docking is use for designing computer assisted drug. Docking is an In-silico method to determine the correct structure of ligand with the target binding site. A basic property of molecular docking is to estimate the predominant attachment site of a ligand with the three dimensional structure of protein using a special scoring feature. The setting up of input for docking is the 3D structure of target proteins and ligands. Docking can be done for virtual screening on large variety of compounds, ranking the result and propose structural hypothesis of ligands inhibiting the target [19]. This new class of small molecular compounds has been shown to have high interaction between target protein and target binding as well as proper absorption, distribution, metabolism and excretion (ADME) to help in target lead selection [20]. Molecular docking also focuses on achieving the system's minimal independent energy, which includes properly aligned proteins and ligands [21].

Small ligands, protein proteins, protein peptides, protein nucleotides can be used in molecular docking. Some of the docking mechanisms are algorithm, ligand flexibility and receptors flexibility. Auto Dock Vina, Auto Dock, CB Dock and some others are mainly the used docking applications [22].

1.2 Problem Statement

Cancer is the second most prevalent cause of death and morbidity in the world. The increase in cancer rate influence the scientist to look for novel drugs obtained from plants sources that have fewer or no side effects as compared to the synthetic drugs. Extracts obtain from plants have been widely used in ethanomedical treatments that have fewer side effects.

1.3 Aims and Objectives

The aim is to identify the novel inhibitors, natural anti-cancer compounds and harmless elements from *Nigella sativa*. Therefore, we focus on protein ligand interactions, which play an important role in structural drug design. To achieve the goal, we have following objectives:

- Identification of various bioactive compounds from Nigella sativa, as potential inhibitors of Anaplastic Lymphoma Kinase (ALK) and Echinoderm Microtubule-Associated Protein-like 4 (EML4).
- 2. To perform the molecular docking mechanism for analyzing the binding confirmation between targeted proteins and ligands.
- 3. To find out the most suitable interacting molecules those have the inhibitory effect against the targeted receptors by physiochemical properties.

1.4 Scope

Cancer is the group of illnesses, which is prevailed in this world and increases the rate of mortality also. It is also very challenging to develop drugs against lung cancer because of the certain reasons like unavailability of complete data, as a result drug formed is not as much effective. Drugs obtained from natural sources are of more worth as compared to the synthetic drugs because of their less toxicity and immunosuppressive activity. Many bioactive compounds are found in the medicinal plants that may prove to be effective for treating such type of cancer. For this purpose, there is a need to identify such compounds that shows inhibitory actions against ALK and EML 4 that could prove beneficial in treating lung cancer.

Chapter 2

Review of Literature

2.1 Cancer

Cancer is meant for a group of illnesses. Any body's cells that start to undergo divisions which are uncontrolled and these cells invade other tissues in the body. The apoptotic process is the programmed cell death mechanism in which cell dies when they become old and perform their assigned task and new cells replace them. This organized process get disrupted by genetic and environmental factors that causes cells to grow and divide indefinitely leading to tumor formation as shown in Figure 2.1. The tumor cells invasion to other nearby tissues is the main cause of mortality and morbidity in cancer patients.

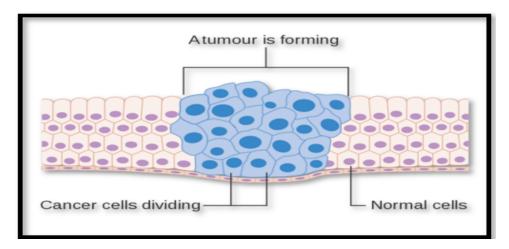


FIGURE 2.1: Tissue showing healthy and cancer cells [24].

2.2 Lung Cancer

Lungs cancer is a form of cancer in which cells in the lungs show uncontrolled division. Lung cancer arises due to uncontrolled division of cells, cause tumor to develop. Lung cancer grows and blocks the airways in the lungs. Fluid accumulates around the lungs pleural space. The tumor may also invade the other parts of body. Lung cancer develops due to change in gene like mutations and environmental factors also contribute in developing lung cancer. Lung cancer mainly diagnosed in older people around the age of 65 and less common in people younger than 45.

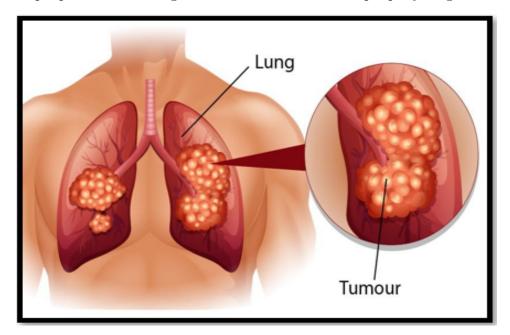


FIGURE 2.2: Tumor shown in Lungs [25].

2.2.1 Lungs Cancer Incidence

Lung cancer is the most common cancer form in US. The incidence varies on the basis of histology. In both males and females, lung cancer is diagnosed. Lung cancer causes more death in males. It reports for 11.6% of the total cases and the mortality rate accounts for 18.4% of the total cancer deaths in 2012. Death rate due to lung cancer caused by passive intake of smoke is 21,400 deaths annually. The leading cause of lung cancer that is smoking attributes to death sentence, ranging more than 80% in United States.

2.2.2 Lungs Cancer Symptoms

Symptoms of Lungs cancer includes:

- Shortness of breath
- Bloody sputum (mucus coughed up from the lungs)
- Fluid in Chest cavity
- Chest pain or discomfort
- Fatigue
- Hoarseness
- Weight loss for not known reason
- Fatigue
- Trouble swelling
- Swelling in the face and/or veins in the neck
- Lack of appetite
- Insomnia [26].

2.2.3 Risk factors for Lungs Cancer

Significant risk factor for Lungs Cancer is increasing age. Other risk factors of Lungs Cancer includes over use of tobacco like use of cigarettes, cigars and pipes.

- Contact to cancer causing substance in passive smoke.
- Experience to different metals like Chromium, Arsenic, Asbestos, Nickle, Beryllium etc.
- Radiation exposure by any of the mean:

- Radiation therapy to the chest.
- Radon exposure at home or at workplace.
- MRI test like computed Tomography (CT) scans [27].
 - 1. Living in area with high air pollution level.
 - 2. Lung Cancer in family history.
 - 3. Human Immunodeficiency Virus (HIV) infection
 - 4. Beta carotene supplements [28].

• Genetics:

Various gene loci are involved in causing the Lungs Cancer. The defects found in growth promoting oncogenes and growth suppressing tumor genes.

- Oncogene KRAS muted in 30% of cases in lung adenocarcinomas [29]
- MYC, Cyclin D1 and EGFR15 are over expressed in 2.5-10%, 5% and
 6% of NSCLC, respectively.
- BCL2 over-articulation is associated with 25% of cases [30].
- BRAF, present in around 2% of adenocarcinoma patients and confined to cancers that didn't show KRAS changes [31].

2.2.4 Lungs Cancer Types

Each type of cancer cells develop and spread in different ways. The most common types of lungs cancer includes following:

- Non-Small Cell Lung Cancer (NSCLC) Non-Small Cell Lung Cancer contribute 80-85% of all lung cancer cases. It further comprises 2 types [32].
- Non squamous carcinoma (includes large-cell carcinoma, adenocarcinoma and other cells types). Adenocarcinoma is the most common type of cancer in non-smokers in United States.
- Squamous carcinoma (include epidermoid carcinoma) [33].

- Small Cell Lung Cancer (SCLC) Small Cell Lung Cancer contributes almost 15 % of the Lung cancer cases. SCLC has more rapid multiplication duration, has high growth fraction and also has an ability to wide spread [34].
- Bronchioloalveolar carcinoma (BAC) It is less common and less prevalent type of cancer. This type of cancer comprises 3-4% of cases, 10-15% of adenocarcinomas having BAC properties [35].

2.2.5 Diagnosis of Lung Cancer

Following are procedures that aid in diagnosing the Lung cancer disease.

- **Sputum Cytology:** Sputum Cytology is the examination of sputum (mucus) that is the fluid secreted by the cells of bronchi and lower respiratory tract. The sputum is examined under microscope to look for cancerous cells.
- Computed Tomography (CT) bronchoscopy: it is the non-invasive technique that shows the internal view of trachea and bronchi in 3D reconstruction [36].
- Flexible bronchoscopy: It is an invasive technique that has been use for diagnostic and for therapeutic processes. It is a safe procedure to diagnose respiratory diseases. It helps in diagnosing patients with infection of chest, parenchymal lung disease, lung nodules, persistent lung infiltrates and lung transplant rejection [37].
- Electromagnetic navigation (EMN) bronchoscopy: It is a procedure that utilizes electromagnetic technique to view and localize endoscopic tools through the bronchial pathway. By using a 3D bronchial map, Physicians are able to locate the desired location in the lungs [38].
- Radial Endobronchial Ultrasound (R-EBUS)-guided Lung Biopsy: For diagnosing peripheral pulmonary malignancies R-EBUS-guided lung biopsy provides a fair diagnostic yield in detecting pulmonary malignancies.

- Transthoracic Needle Aspiration (TTNA): It is an accurate modality that is used for biopsy of lung pathology. It directs the biopsy tools toward the area of abnormality [39].
- Pleural Biopsy: Pleural biopsy is a minimal invasive procedure that is much sensitive and specific. It diagnoses pleural diseases. A small piece is taken by using the special biopsy needle to look for infection, cancer or any other disease condition [40].

2.2.6 Lung Cancer Treatments

Surgery: Surgery entails the physical removal of the tumor as well as any of the underlying tissue that cause spread of disease. Certain procedure involve in treating lungs cancer are:

Lobectomy: The affected lobe of lung is removed by applying this procedure. It is the most commonly performed surgical treatment against lung cancer.

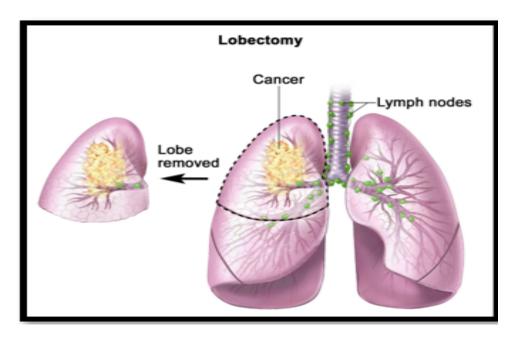


FIGURE 2.3: Removal of affected part by Lobectomy.

Segmentectomy: Each of the lungs comprises of two to five segments and surgeons removed that segments which are infected leaving the uninfected one.

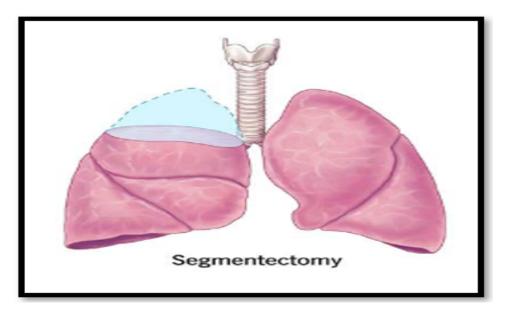


FIGURE 2.4: Removal of infected part by Segmentectomy.

Wedge Resection: The Removal of small wedge shape part of lung that surrounds the tumor.

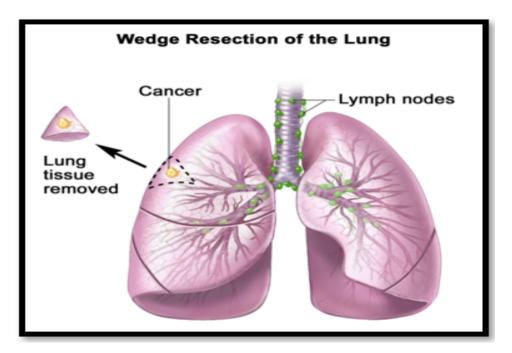


FIGURE 2.5: Removal of wedge shape part of lung by Wedge Resection.

Pneumonectomy: The removal of whole lung affected by the cancer. This procedure is mainly done when the cancer cannot be removed easily by lobectomy

[41].

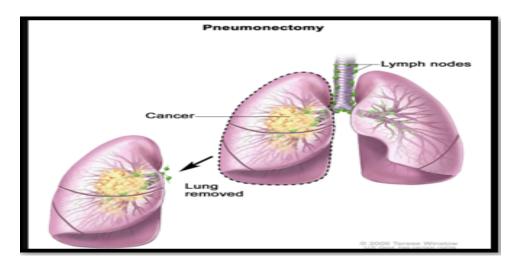


FIGURE 2.6: Removal of whole lung by Pnuemonectomy.

Radiation Therapy: It involves the use of high frequency rays including Xrays or Gamma rays to treat a cancer or post-surgery tumor site. The rays are strong enough to destroy the cancer cells that can recruit growth where they were removed. Treatments are usually issued five days a week for duration of five to seven months. Each treatment takes approximately 15 minutes [42].

Chemotherapy: It is a form of drug therapy that is used to destroy cancerous cells. This therapy can be used alone or sometimes it can be used in combination with other clinical treatments like surgery. Most of the commonly used medicines for non-small cell lung cancer are Crizotinib, Sunitinibmalate, and Tandutinib. Chemotherapy has many unfavorable ill effects, so there is a need to consult the doctor before the start of chemotherapy [43].

2.3 Medicinal Plants

Usage of medicinal plants for the treatment purpose is very old. In the ancient time, due to less knowledge, people firstly use the herb and afterward get knowledge about its efficacy. Awareness about the use of medicinal plants is the result of many years of man struggles for their discovery, their usage and their beneficial effects. Contemporary science has recognized their dynamic activity that wide range of drugs that have plant origin has been known by the ancient civilization. The development in the ideas related to medicinal plants as well as increased awareness make physicians and scientists to respond to the emerging situations in health sector [44]. The word 'Medicinal Plants refers to a variety of plants used in herbal medicines. It is the practice of using plants for medicinal purposes as well as for study purpose. Medicinal plants contain the rich source of ingredients that can be used to make pharmacopoeia, non- pharmacopoeia and synthetic drugs. Since ancient time, medicinal plants and aromatic plants have been used as therapeutic agents [45]. Natural source drugs accounts for about 40% of newly approved drugs in the last two decades. They play an important role in the discovery of drugs for cancer treatment and dealing with other infectious ailments [46]. In ancient medicinal systems, such as Ayurvedic, Unani and Chinese traditional medicines, hers has been used to treat different kinds of diseases and infections [47].

2.4 Natural Compounds Targeting Cells of Lungs Cancer

Natural products contain lead compounds that have been used for targeting cells of Lung Cancer. Some of these are mentioned in the Table 2.1 given below

Structure	Source	Protein/signal
		DNA damage or
		repair,
		JAK-STAT,
	Cucurma longa	Sonic Hedgehog,
		CD133, CD44,
		ALDHA1, Nanog,
		Oct4
	Structure	Structure Source

TABLE 2.1: Natural products targeting lung cancer cells [48].

Compounds	Structure	Source	Protein/signal
			AKT,
Gigantol	H,COOCH_	Dendribium draconis	CD133,
			ALDH1A1
			OCT-4,
Salinomycin	CONTRACTION OF CONTRACTICON OF CONTRA	Streptomyces albus	Nanog,
	44d ⁻		Sox2
Silibinin		Silybum mariamum	ALDH
SIIDIIII	HO, A. JO, LLLL, HO	Suyoum mariamum	activity
			AKT-
			proteasomal
	O L		degradation,
Vanilin	HO OCH3	Vanilla planifolia	CD133,
			ALDH1A1,
			Oct4,
			Nanog
	QCH ₅		CD133,
Renieranycin		Xestospongia sp	CD44,
	CH3		ALDH1A1
			ER stress,
	н		Apoptosis,
Parthenolide		Tanacetum parthenium	ATF4,
			DDIT3,
			PMAIP

TABLE 2.1: Natural products targeting lung cancer cells [48].

2.5 Nigella sativa

Nigella belongs to family Ranunculaceae is a small genus comprising around 20 species. The plants belong to this genus are annuals and get through unfavorable

condition as seeds. The seeds are bifurcate or discoid shape and have characteristics black color that's why they are commonly called "Black cumin" [49]. This genus produces secondary metabolites and essential oils that are used to treat various diseases. *Nigella spp.* has a strong healing powers and food importance. As well as *Nigella* seeds are rich in Linoleic acid and Omerga-6 fatty acid and provide dietary phytochemicals including thymoquinone, saponins, flavonoids and alkaloids [50]. The evolutionary origin of Nigella species are the Western-Irano-Turanian region. This Species is found in Pakistan, India, Bangladesh, Russia, Southern Europe, North Africa, Turkey and Middle-East [51].

Nigella sativa is widely used in ancient times for treating lung cancer, diabetes, high BP, respiratory pathologies, allergy and hyper-sensibility. It also acts as antirheumatic and analgesic (pain reliever). N. sativa is also use as food preservative and as spice [52]. The seeds of N. sativa contain very low degree of toxicity. The extract of seed has a property against nephrotoxicity and hepatotoxicity. The other functions performed by the N. sativa are as antimicrobial, antipyretic, antiinflammatory and antineoplastic [53].



FIGURE 2.7: The figure represents Nigella sativa flower [54].

2.5.1 Taxonomic Hierarchy

Nigella sativa is the binomial name of the plant belonging to family Ranunculaceae. They are easily grown in loamy soil. The plants grow best in full sunlight however it also show tolerance in shady places. It is widely distributed in different regions of the world. The taxonomic hierarchy is shown in Table 2.2.

Sr no.	Domain	Eukarya
1-	Kingdom	Plantae
2-	Subkindgom	Tracheobionta (Vascular plants)
3-	Division	Spermatophyta (Seeded plants)
4-	Class	Magnoliphyta (Flowering plants)
5-	Subclass	Magnoliidae
6-	Order	Ranunculales
7-	Family	Ranunculaceae
8-	Genus	Nigella
9-	Species	Nigella sativa

TABLE 2.2: Taxonomic hierarchy of Nigella sativa [55].

2.6 Anti-Cancer Mechanism of Action of Bioactive Constituents of Nigella species

N. sativa contains bioactive like Isoquinoline, Beta-pinene, Apigenin, Salfredin B11, Pyrazole, Pyragallol, Salicylic acid. *N. sativa* exhibit anticancer effect due

to the presence of TQ, which is extracted from the seeds of N.sativa. Concentration of 100µm is significant enough to inhibit cancer cells growth by about 90% [56]. Essential oils are also the part of bioactive compounds like linoleic acid.

2.7 Targeted Proteins

There are 2 different types of proteins which are used as the targeted protein for molecular docking process such as Anaplastic Lymphoma kinase and Echinoderm Microtubule-associated Protein-Like 4.

2.7.1 EML 4 and ALK Fusion Oncogenes

Lung cancer development is linked to the EML4 and AKL fusion oncogenes that arise from the inversion on chromosome 2. In adult tissues, ALK fuse with the EML4 results in the formation of tumor in the lungs. ALK is a receptor of tyrosine kinase. The fusion of ALK to EML4 does not always occur in the same location rather it gets change, as a result give rise to multiple variants [57]. EML4-ALK fusion proteins go through the process of dimerization which is ligand independent. This type of fusion including other types has been seen in NSCLC. The chromosomal inversion does not usually arise within the identical precise location, however, giving upward push to more than one ALK-EML4 forms, five all of which incorporate the equal intracellular tyrosine kinase domain of ALK but exclusive truncations of EML4. Fusion of ALK with other genes like TFG and KIF5B has also been diagnosed in NSCLC, although those fusions seem like lots less commonplace than ALK-EML4. ALK fusion proteins, which include ALK-EML4, undergo ligand-impartial dimerization mediated by way of the coiled-coil domain of the fusion accomplice, resulting in constitutive activation of the ALK tyrosine kinase. The ALK-EML4 fusion protein uncovers marked remodeling hobby both in vitro and in vivo. In a transgenic mouse model, lung-precise expression of ALK-EML4 outcomes accordingly within the development of lung adenoma.

Chapter 3

Research Methodology

3.1 Context Diagram

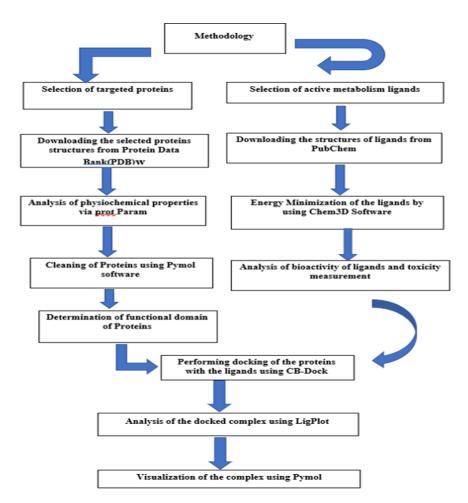


FIGURE 3.1: The methodology flow chart.

3.2 Selection of Disease

Cancer can be described as the out of control boom of extraordinary cells. Most lungs cancer is non-small cell lung cancer (NSCLC). Oncogenic fusion genes together with ALK and EML4 are found in non–small–cell lung cancers, act for 2 to 7% of such tumors. ALK proteins play a critical function in deactivating the apoptosis system in cancer. Non–small cell lung cancer cells need ALK to mobile increase and proliferation of the cells in lungs cancer [59].

3.3 Selection of Proteins

Structure of ALK and EML4 were taken from protein databank. The PDB archive is the only site where you can learn about the three dimensional structure of large biological molecules like proteins and nucleic acid [60].

3.4 Primary Sequence Retrieval

The primary sequence of target proteins (ALK and EML4) were obtained in FASTA format from UniProt Database (https://www.uniprot.org) [61].

3.5 Analysis of Physiochemical Properties

The function of proteins is primarily determined by their physiochemical properties. These properties were predicted by using ProtParam.

Physiochemical parameters were investigated using the ProtParam tool of Expasy including molecular weight number of amino acids, iso electric point, instability index, grand average of hydropathicity (GRAVY), number of negatively charge residues, number of positively charge residues, aliphatic index and amino acids and atomic composition [62].

3.6 Cleaning of the Downloaded Proteins

The extra constituents attached to the proteins must be extracted after downloading the proteins structures, which was achieved using the open source system PyMol [63].

3.7 Determination of Functional Domains of Target Proteins

InterPro, a database that can analyze a protein and provide information about the families, functional sites and domains of the proteins under study, is used to determine the domains of the target proteins. The polypeptide binding sites and homo dimer interfaces were obtained by inserting the receptor proteins FASTA sequence.

3.8 Selection of Active Metabolic Ligands

The ligands that have already shown antiviral, antioxidants and antimalarial properties were chosen. Sterols, Phenolic compounds, Terpenes, monoterpenes, flavonoids, sesquiterpenes, and ad monoterpenes are among them [64].

3.9 Ligands Preparation

PubChem database was used to download the three dimensional structures of all the above ligands. PubChem is a database that contains information about the chemical molecules and is run by the National Center for Biotechnology Information (NCBI). The data is correlated with chemical names and molecular formulas, 3D or basic structures, their isomers and canonic structures. The structures of the ligands which were obtained from PubChem was downloaded and then the ligands MM2 energy was minimized by using Chem3D ultra. At the end, SDF format was select to save the energy minimized structures of the ligands [65].

3.10 Molecular Docking

For the purpose of interpreting docking effects the interaction of the ligands active pockets with the protein were measure. Ionic bonding, Hydrogen Bonding and hydrophobic bonding are the three types of interactions investigated. CB-Dock were used to performed the molecular docking between the protein and the ligand. CD-Dock automatically locates docking positions. CB Dock is a docking method for proteins and ligands that measures the bonding sites, their duration, and their center. The size of the box is changed to match the ligand and then docking is completed. Since the docking is based on cavity binding, the accuracy ratio is higher. 3D structures of the protein in PDB format and the 3D structure of the ligand in SDF format were uploaded to perform the docking [66].

3.11 Visualization of Docking Results via PyMol

Over the past few years, the PyMol has emerged as an efficient molecular tool of visualization. The graphics and its ability to view 3D structures have been extra ordinary [67]. PyMol provides a plug in which can access the results and make their visualization clearer so that the docking results can be studied easily. The pictures of the docking results were captured also. The docking results were saved in PDB format throughout the process.

3.12 Analysis of Dock Complexes via LigPlot

Once the dock complexes were obtained with lowest Vina score, the analysis of docking complex were the next step. The complexes were stored as a PDB files the program LigPlot were used to perform the research. The schematic diagram of the proteins and ligand interactions were created automatically for the given PDB file format. The hydrophobic and the hydrogen bonding interactions were studied using LigPlot. LigPlot provides a 2D representation of the protein ligands complex using this tool [68].

3.13 Selection of Standard Drug against Lungs Cancer

Standard drugs against lung cancer were selected based on docking values, physiochemical properties and ADMET properties.

3.14 Ligands ADME Poperties

Is in general, a more effective drug discovery needs a lead is more like the drug. The compounds was then tested for the drug score, drug similarity and toxicity. The ADME or Absorption, Distribution, Metabolism and Excretion of the human body can be optimized using the pkCSM [69].

3.15 Lead Compounds Analysis and Toxicity Measurements

The most active inhibitors were discovered after a careful study of proteins and ligands interactions, docking ratings and toxicity studies. Our lead compounds are the one we've choose. After applying the rules of 5, the lead compound is defined.

- 1. The log value of the drug like compound must be limited to five
- 2. The molecular weight of the compound must be less than 500

- 3. Hydrogen bond acceptors number must be ten.
- 4. Hydrogen bond donors number must be 5

Once any compound fit these rules, it is selected as lead compound.

3.16 Comparison of Standard Drugs and Lead Compounds

The comparison between standard anticancer drugs and the proposed lead compounds were done through comparing docking values, physiochemical properties and ADMET properties [70].

Chapter 4

Results and Discussions

4.1 Structure Modeling

4.1.1 Primary Sequence Retrieval

Primary sequences of target proteins (ALK and EML4) were taken in FASTA format from Uniprot database (http://www.uniprot.org) under accession number of Q9UM73 for ALK and P68363-P07437 for EML4 chains.

>sp—Q9UM73—ALK-HUMAN ALK tyrosine kinase receptor OS=Homo sapiens OX=9606 GN=ALK PE=1 SV=3

MGAIGLLWLLPLLLSTAAVGSGMGTGQRAGSPAAGPPLQPREPLSYSRLQR VDFVVPSLFRVYARDLLLPPSSSELKAGRPEARGSLALDCAPLLRLLGPAP TAGSPAPAEARTLSRVLKGGSVRKLRRAKQLVLELGEEAILEGCVGPPGEA LQFNLSELFSWWIRQGEGRLRIRLMPEKKASEVGREGRLSAAIRASQPRLL GTGHSSLESPTNMPSPSPDYFTWNLTWIMKDSFPFLSHRSRYGLECSFDFP YSPPLHDLRNQSWSWRRIPSEEASQMDLLDGPGAERSKEMPRGSFLLLNTS HTILSPWMRSSSEHCTLAVSVHRHLQPSGRYIAQLLPHNEAAREILLMPTP WTVLQGRIGRPDNPFRVALEYISSGNRSLSAVDFFALKNCSEGTSPGSKMA FTCWNGTVLQLGQACDFHQDCAQGEDESQMCRKLPVGFYCNFEDGF GLPLEAATAPGAGHYEDTILKSKNSMNQPGP >sp—P68363—TBA1B-HUMAN OS=Homo sapiens OX=9606 GN=TUBA1B PE = 1 SV=1

MRECISIHVGQAGVQIGNACWELYCLEHGIQPDGQMPSDKTIGGGDDSF TGAGKHVPRAVFVDLEPTVIDEVRTGTYRQLFHPEQLITGKEDAANNYA GKEIIDLVLDRIRKLADQCTGLQGFLVFHSFGGGTGSGFSLLMERLSVD LEFSIYPAPQVSTAVVEPYNSILTTHTTLEHSDCAFMVDNEAIYDICRR PTYTNLNRLISQIVSSITASLRFDGALNVDLTEFQTNLVPYPRIHFPLA SAEKAYHEQLSVAEITNACFEPANQMVKCDPRHGKYMACCLLYRGDVVPK NAA IATIKTKRSIQFVDWCPTGFKVGINYQPPTVVPGGDLAKVQRAVCML-SNTTAIAEA WARLDHKFDLMYAKRAFVHWYVG EGMEEGEFSEARED MAALEKDYEEVGVDSVEG EGEEEGEEY

>sp—P07437—TBB5-HUMAN OS=Homo sapiens OX=9606 GN=TUBB PE=1 SV=2

MREIVHIQAGQCGNQIGAKFWEVISDEHGIDPTGTYHGDSDLQLDRISVYYNEA TGG KYVPRAILVDLEPGTMDSVRSGPFGQIFRPDNFVFGQSGAGNNWAKGH-GAELVD SVLDVVRKEAESCDCLQGFQLTHSLGGGTGSGMGTLL ISKIREEYPDRIMNTFSVVP SPKVSDTVVEPYNATLSVHQLVENTDETYCID-NEALYDICFRTLKLTTPTYGDLNHL VSATMSGVTTCLRFPGQLNAMVPFPRL-HFFMPGFAPLTSRGSQQYRALT VPELTQQVFDAKNMMAACDPRHAAVFR-GRMSMKEVDEQMLNVQNKNSSYFV EWIPNNVKTAVCDIPPRGLKMAVTFI GNTEAESNMNDLVSEYQQYQDATAEEEEDFGEEAEEEAMFRRKAFLHWYT STAIQELFKRISEQFTAMFRRKAFLHWYT GEGMDEEFTEAESNMNDLVSEY QQYQDATAEEEEDFGEEAEEEA

Anaplastic Lymphoma Kinase and Echinoderm Microtubule associated Protein like 4 were selected as the target proteins and Isoquinoline, β -pinene, apigenin, Salfredin B11, pyrazole, pyragallol, salicylic acid, syringic acid, gallic acid, camphene, 3,4-dihydroxybenzoic acid, 4-dihydroxycinnamic acid, caffeic acid, myristic acid and stearic acid were selected as ligands in this research work.

4.1.2 Physiochemical Characterization of ALK and EML 4

ProtParam is a tool of Expasy which is used online for the calculation of various physical and chemical parameters for ALK and EML4 proteins stored in Swissprot or TrEMBL or for a user entered protein sequence. The estimated values of following parameters includes the molecular weight, amino acid composition, theoretical pI, atomic composition, extinction coefficient, estimated half-life, instability index, aliphatic index and grand average of hydropathicity (GRAVY). The calculated pI greater than 7 represents the basic nature of the protein while less than 7 show acidic nature of protein. Extinction coefficient represents light absorption. In stability index if less than 40 shows stability of the protein while greater than 40 indicates the instability of the protein [71].

The table 4.1 and 4.2 shows the physiochemical properties of ALK and EML4 respectively.

PROPERTIES	VALUES	
Molecular	174606.62	
Weight	Da	
Theoretical	6.68	
pI	0.08	
Negative- charged	161	
Residues	101	
Positive-charged	154	
Residues	154	
Extinction	233505	
Coefficient 1	M-1 cm-1	
Extinction	231130	
Coefficient	M-1	
2	cm-1	

TABLE 4.1: Physiochemical Properties of ALK.

PROPERTIES	VALUES
Instability	51.29
index	01.29
Aliphatic	77.08
index	11.00
GRAVY	-0.322

 TABLE 4.1: Physiochemical Properties of ALK.

TABLE 4.2: Physiochemical Properties of EML4.

PROPERTIES	VALUES
Molecular	108916.22
Weight	Da
Theoretical pI	5.96
Negative- charged Residues	129
Positive-charged	109
Residues	
Extinction	135885
Coefficient 1	M-1 cm-1
Extinction	134760
Coefficient 2	M-1 cm-1
Instability	
	37.32
index	
Aliphatic	
	75.31
index	
GRAVY	-0.538

The protein's aliphatic composition is indicated by the aliphatic index. The high value of the aliphatic index indicates the thermo stability of the protein. Molecular weight includes both positive and negative charged residues of the protein. At 280nm the ranging extinction coefficient of 73980, 67965, 20105 and 112270 indicates Tyr and Trp high concentration [72].

Low GRAVY shows better interaction with water molecules. All these parameters which are selected for this research work are taken according to previous research work. MW stands for Molecular Weight pI for theoretical isoelectric point (pH at which protein is neutral, without any charge), NR for total negatively charged residues (Asp+ Glu), PR for total positively charged residues (Arg + Lys), Ext.Co1for extinction coefficients when assuming all pairs of Cys residues from cysteine, Ext.Co2 for extinction coefficients when assuming all Cys residues are reduced, and GRAVY for grand average of hydropathicity.

4.1.3 3D Structure Predictions of Proteins

3D Structure of targeted proteins was downloaded from RCSB PDB in PDB format. Protein Data Bank is a three dimensional database of complex molecules of living organisms, like nucleic acids and proteins. The 3D Structures of ALK and EML4 were obtained from PDB named as 2XB7 and 6I2I under accession number 10.2210/pdb2XB7/pdb and 10.2210/pdb4CGC/pdb respectively 4.1, 4.2.

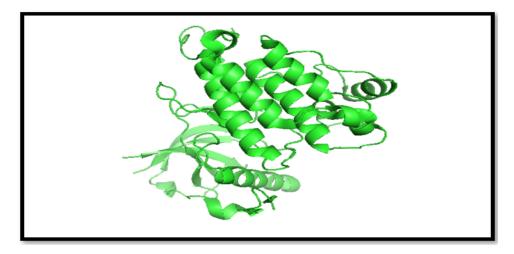


FIGURE 4.1: Human Anaplastic Lymphoma Kinase

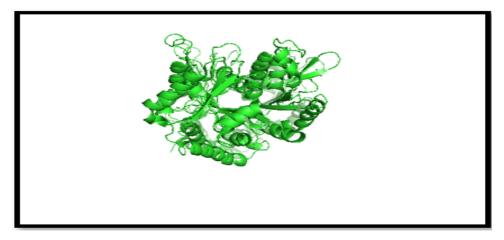


FIGURE 4.2: Human Echinoderm microtubule associated Protein Like-4

4.1.4 Functional Domain Identification of Proteins

Functional domain is the active part of protein that is involve in interactions of protein with other substances. Protein can have more than one functional domain that performs different functions [73].

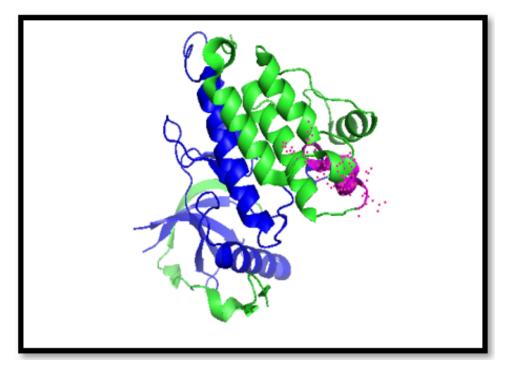


FIGURE 4.3: Functional domains of Human ALK

Figure 4.3 shows functional domains of human ALK protein. ALK has many like MAM domain (264-427), MAM domain 478-636, Ser-Thr/Tyr-kinase-cat-domain (1117-1383), Prot-kinase-domain (1116-1392).

Figure 4.4 shows functional domains of human EML4 protein. EML4 has Tubulin -layer-sand-domain (248-393), Tubulin-FtsZ-GTPase domain (3-246) domains.

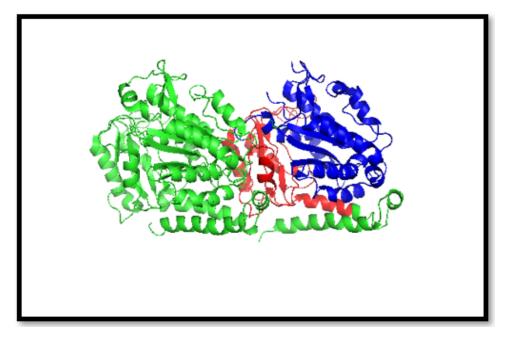


FIGURE 4.4: Functional domains of Human EML4

4.1.5 Templates Selection

The 3D structure of the selected templates were taken from the protein data bank (PDB) and listed in 4.3.

Templates	Resolution	PDB ID	Structures
Human Anaplastic Lymphoma Kinase Human	2.50 Å	2XB7	
Echinoderm Microtubule Associated Protein Like-4	3.60 Å	6I2I	

TABLE 4.3 :	Selected	PDB	Templates	Structures.
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4.2 Structures of Proteins Refined for Docking

The selected 3D structures were refined by PyMol for docking and are shown in Figures 4.5 and 4.6.

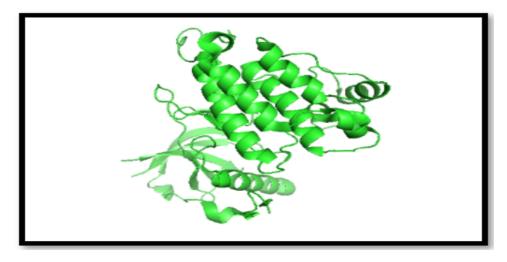


FIGURE 4.5: Refined 3D Structure of Human ALK

Figure 4.5 shows the refined structure of protein ALK. The tool PyMol removes any additional ligand and the water droplets. Now the protein is ready for docking purpose.

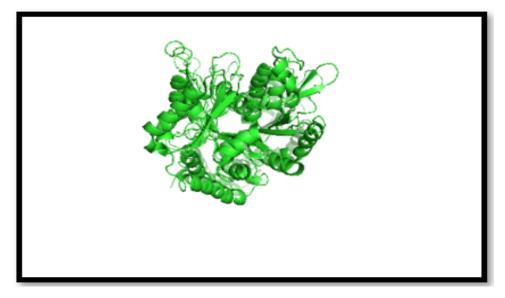


FIGURE 4.6: Refined 3D Structure of Human EML-4

Figure 4.6 shows the refined structure of protein EML4. The tool PyMol removes any additional ligand and the water droplets. Now the protein is ready for docking purpose.

4.3 Ligands Selection

Protein data bank contains a large amount of protein ligand complex, especially for the protein target. Therefore, the selection of ligands is based on the best resolution of the structure, the chemical class of the co-crystal ligand bound to the protein structure and the best binding affinity. Conformational selection is a process in which ligand selectively binds to one of these conformers, strengthening it and increasing its population with respect to the total population of the protein is ultimately resulting in the final observed complex. Ligands (compounds of the selected plant) were searched out from PubChem, which is the world's largest freely accessible chemical information database. Their 3-D structures were downloaded from PubChem in SDF format. Selected compounds were representing all the classes of compounds like Alkaloids, essential oils etc. After selection of ligands, energy minimization of ligands was done which was carried out by Chem pro software (Chem 3D v 12.0.2). This was a mandatory step in the preparation of ligands for docking because unstable ligands will show unreliable vina scores in docking results. Bioactive compounds of Nigella sativa were selected as ligands for the present study (Table 4.4). The 3D structures and information of selected ligands that were Isoquinoline, β -pinene, apigenin, salfredin B11, pyrazole, pyragallol, salicylic acid, syringic acid, gallic acid, camphene, 3,4-dihydroxybenzoic acid, 4-dihydroxycinnamic acid, caffeic acid, myristic acid and stearic acid were downloaded from PubChem. This database (http://pubchem.ncbi.nih.gov) is a public repository for information on chemical substances and their biological activities [74]. 4.4 shows the selected Ligands with their structural information.

Name	Molecular Formula	Molecular Weight	Structure
Isoquinoline	$\rm C_9H_7N$	129.16 g/mol	X\$\$.

TABLE 4.4: Selected Ligands with Structural Information

β -pinene	$\mathrm{C_{10}H_{16}}$	136.23 g/mol	
Apigenin	$\mathrm{C}_{15}\mathrm{H}_{10}\mathrm{O}_{5}$	270.24 g/mol	\$245
Salfredin B11	$C_{10}H_{12}O_4$	232.23 g/mol	-2025
Pyrazole	$\mathrm{C_{3}H_{4}N_{2}}$	68.03 g/mol	X.
Pyragallol	$C_6H_6O_3$	126.11 g/mol	- Ar
Salicylic Acid	$\mathrm{C_7H_6O_3}$	138.12 g/mol	-\$ 7 4.
Syringic Acid	$\mathrm{C_9H_{10}O_5}$	198.17 g/mol	2 Č
Gallic Acid	$C_7H_6O_5$	170.12 g/mol	24
Camphene	$\mathrm{C}_{10}\mathrm{H}_{16}$	136.23 g/mol	
3,4-dihydroxy- benzoic Acid	$\mathrm{C_7H_6O_4}$	154.12 g/mol	574
4-dihydroxy- cinnamic Acid	$C_9H_8O_3$	164.16 g/mol	-7574
Caffeic Acid	$\mathrm{C}_{9}\mathrm{H}_{8}\mathrm{O}_{4}$	180.16 g/mol	2444
Myristic Acid	$\mathrm{C}_{14}\mathrm{H}_{28}\mathrm{O}_2$	228.37 g/mol	
Stearic Acid	$C_{18}H_{36}O_2$	284.5 g/mol	dissing the

4.4 Virtual Screening and Toxicity Prediction

Drug like compounds are separated from non-drug like compounds by following certain parameters like Lipinski's rule of five and ADMET properties test [75]. The original rule of five deals with four physiochemical parameters (Molecular weight ≤ 500 , log P value ≤ 5 , H-bond ≤ 5 , and H-bond acceptors ≤ 10) that is associated with orally active compounds [76]. A compound considered as drug likeness if it is complying with three or more of the RO5. If a compound violates more than two of these rules, it is assumed to be poorly absorbed [76]. 4.5 showed the applicability of Lipinski's rule of five on selected ligands. All ligands follow these rules.

Ligand	Log	Molecular	H-Bond	H-Bond	
Ligaliu	Р	Weight	Acceptor	Donor	
Isoquinoline	2.	129.16	1	0	
isoquinonne	2348	g/mol	T	U	
β -pinene	2.	136.23	0	0	
<i>p</i> -pmene	9987	g/mol	0	0	
Apigenin	2.	270.24	5	3	
rpigeinn	5768	g/mol	0	5	
Salfredin	2.	232.23	4	1	
B11	2468	g/mol	4	1	
Pyrazole	0.	68.03	1	1	
I ylazole	4097	g/mol	1		
Pyragallol	0.	126.11	3	3	
i yraganor	8034	g/mol	0		
Salicylic	1.	138.12	2	2	
acid	0904	g/mol	2	2	
Syringic	1.	198.17			
			4	2	
acid	1076	g/mol			

TABLE 4.5: Applicability of Lipinski's Rule on Ligands

Ligand	Log	Molecular	H-Bond	H-Bond	
Liganu	Р	Weight	Acceptor	Donor	
Gallic	0.	170.12	4	4	
acid	5016	g/mol	4	'T	
Camphene	2.	136.23	0	0	
Camphene	997	g/mol	0	0	
3,4-dihydroxy-					
benzoic	0.	154.12	3	3	
	796	g/mol	ა	5	
acid					
4-dihydroxy-					
cinnamic	1.	164.16	2	0	
	49	g/mol	Ζ	2	
acid					
Caffeic acid	1.	180.16	3	3	
Callele acid	1956	g/mol	ა		
Memiatio acid	4.	228.37	1	1	
Myristic acid	7721	g/mol	Ţ	1	
Steenie:-]	6.	284.5	1	1	
Stearic acid	335	g/mol	1	1	

TABLE 4.5: Applicability of Lipinski's Rule on Ligands

4.4.1 Toxicity Prediction

pkCSM is an online tool used to find the ADMET (Absorption, Distribution, Metabolism, Excretion and Toxicity) properties of bioactive compounds and drugs. By using this tool, we will determine the toxicity of selected ligands.

AMES toxicity test is used to test the mutagenic potential of the compound by using bacteria. If it shows a positive response, then ligand is mutagenic which act as a carcinogen. The maximum tolerated dose (MRTD) provides a measure of toxic chemical limits on individuals. This will help in directing the first recommended dose of the treatment regimen in phase 1 clinical trials. MRTD is expressed in the form of logarithms (log mg/kg/day). For a specific compound, MRTD is higher if it is greater than 0.477log (mg/kg/day), and lower if it is less than 0.477log (mg/kg/day). The hERGI and II inhibitors model determine the potential of any compound to cause the inhibition of potassium channels induced by the hERG (human ether-a-go-go gene). An inhibitor of these channels could probably lead to the chronic QT syndrome and a long term basis the person could develop fatal ventricular arrhythmia. Many useful products from the pharmaceutical market have been removed as a result of hERG channel inhibitor.

LD50 is the quantity of a compound that causes the deaths of 50% of experimental animals (mice). The LD50 (mol/kg) predicts toxicity of a probable compounds whereas LOAEL aims to identify the lowest dosage of a compound with a significant adverse effect. Exposure to low to moderate chemical dose for a long time is very important in medicine and is expressed in a log (mg/kg-bw/day). Hepatotoxicity determines liver damage that is induced by drug and is a major safety issue for drug development. Skin sensitization is a potential negative effect of skin care products. *T. pyriformis* is protozoan bacteria which toxin is often used as toxic endpoint (IGC50) and inhibits 50% growth. *T.pyriformis* IGC50 (negative concentration logarithm required to prevent 50% growth) in log ug/L predicted value >-0.5 log ug/L is considered toxic. The lethal concentration (LC50) represents the concentration of molecules needed to cause the death of 50% of Flathead Minnows (small bait fishes). In Minnow toxicity LC50 values below 0.5mM (log LC50 <-0.3) are regarded as high acute toxicity.

4.4.2 Toxicity Predicted Values of Selected Ligands

4.4.2.1 Isoquinoline

Isoquinoline is non-carcinogenic and it shows high Max.tolerated dose. Maximum tolerated dose helps in deciding maximum starting dose in phase I of clinical trial. It is supporter of potassium channels and is non-hepatotoxic. This ligand is skin sensitive. *T. pyriformis* and Minnow toxicity are within recommended range.

4.4.2.2 β -pinene

 β -pinene is non-carcinogenic and it shows high Max.tolerated dose. It is supporter of potassium channels and is non-hepatotoxic. *T. pyriformis* showed high value of toxicity. Minnow toxicity is within recommended range.

4.4.2.3 Apigenin

Apigenin is non-carcinogenic and it shows low Max.tolerated dose. It is supporter of potassium channels and is non-hepatotoxic. *T. pyriformis* and Minnow toxicity are also within recommended range.

4.4.2.4 Salfredin B11

Salfredin B11 is non-carcinogenic and it shows low Max.tolerated dose. It is supporter of potassium channels and is non-hepatotoxic. *T. pyriformis* and Minnow toxicity are also within recommended range. This ligand shows all the predicted value within the safe range.

4.4.2.5 Pyrazole

Pyrazole is non-carcinogenic and it shows high Max.tolerated dose. All the toxicity value are within recommended range.

4.4.2.6 Pyragallol

Pyragallol is non-carcinogenic and it shows low Max.tolerated dose. It is supporter of potassium channels and is non-hepatotoxic. *T. pyriformis* and Minnow toxicity are also within recommended range.

4.4.2.7 Salicylic Acid

Salicylic acid is non-carcinogenic and it shows high Max.tolerated doses. It is supporter of potassium channels and is non-hepatotoxic. *T. pyriformis* and Minnow toxicity are also within recommended range.

4.4.2.8 Syringic Acid

Syringic acid is non-carcinogenic and it shows high Max.tolerated doses. It is supporter of potassium channels and is non-hepatotoxic. *T. pyriformis* and Minnow toxicity are also within recommended range. This ligand shows all the predicted value within the safe range.

4.4.2.9 Gallic Acid

Gallic acid is non-carcinogenic and it shows high Max.tolerated dose. It is supporter of potassium channels and is non-hepatotoxic. *T.pyriformis* and Minnow toxicity are also within recommended range. This ligand shows all the predicted value within the safe range.

4.4.2.10 Camphene

Camphene is non-carcinogenic and it shows low Max.tolerated doses. It is supporter of potassium channels and is non-hepatotoxic. *T. pyriformis* value show slightly high range. Minnow toxicity is also within recommended range. This ligand shows all the other predicted value within the safe range.

4.4.2.11 3,4-dihydrobenzoic Acid

3,4-dihydrobenzoic acid is non-carcinogenic and it shows high Max.tolerated doses. It is supporter of potassium channels and is non-hepatotoxic. *T. pyriformis* and Minnow toxicity are also within recommended range.

4.4.2.12 4-dihydroxycinnamic Acid

4-dihydroxycinnamic acid is non-carcinogenic and it shows high Max.tolerated dose. It is supporter of potassium channels and is non-hepatotoxic. *T. pyriformis* and Minnow toxicity are also within recommended range. This ligand shows all the other predicted value within the safe range.

4.4.2.13 Caffeic Acid

Caffeic acid is non-carcinogenic and it shows high Max.tolerated dose. It is supporter of potassium channels and is non-hepatotoxic. *T. pyriformis* and Minnow toxicity are also within recommended range. This ligand shows all the predicted value within the safe range.

4.4.2.14 Myristic Acid

Myristic acid is non-carcinogenic and it shows low Max.tolerated dose. It is supporter of potassium channels and is non-hepatotoxic. Myristic acid shows skin sensitization. *T. pyriformis* and Minnow toxicity values show high toxicity range.

4.4.2.15 Stearic Acid

Stearic acid is non-carcinogenic and it shows low Max.tolerated doses. It is supporter of potassium channels and is non-hepatotoxic. Stearic acid shows skin sensitization. *T. pyriformis* and Minnow toxicity values show high toxicity range.

Toxicity predicted values of selected ligands are listed in 4.6 and 4.7. This table include the

1. Ligands, AMES toxicity, max. tolerated dose, hERG I inhibitor, hERG II inhibitor, oral rat acute toxicity, oral chronic toxicity,

2. Hepatotoxicity, skin sensitization, *T. pyriformis* toxicity and minnow toxicity.

Ligands	AMES Toxicity	Max. Tolerated Dose- Human (mg/kg)	hERG I Inhibitor	hERG II Inhibitor	Oral Rat Acute Toxicity (mol/kg)
Isoquinoline	No	0.694	No	No	2.216
β -pinene	No	0.371	No	No	1.673
Apigenin	No	0.328	No	No	2.45
Salfredin B11	No	-0.051	No	No	1.701
Pyrazole	No	0.818	No	No	2.186
Pyragallol	No	-0.269	No	No	2.049
Salicylic Acid	No	0.61	No	No	2.282
Syringic Acid	No	1.374	No	No	2.157
Gallic Acid	No	0.7	No	No	2.218
Camphene	No	0.305	No	No	1.554
3,4-Dihydroxy- benzoic Acid	No	0.814	No	No	2.423
4-Dihydroxy- cinnamic Acid	No	1.111	No	No	2.155
Caffeic Acid	No	1.145	No	No	2.383

TABLE 4.6: 4.6a)Toxicity predicted values of Ligands

Ligands	AMES Toxicity	Max. Tolerated Dose- Human (mg/kg)	hERG I Inhibitor	hERG II Inhibitor	Oral Rat Acute Toxicity (mol/kg)
Myristic Acid	No	-0.559	No	No	1.477
Stearic					
Acid	No	-0.791	No	No	1.406

TABLE 4.6: 4.6a)Toxicity predicted values of Ligands

TABLE 4.7: b). Toxicity predicted values of Ligands

Ligands	Oral Rat Chronic	Hepato-	Skin Sensit-	T. pyriformis	Minnow toxicity	
Liganus	Toxicity	toxicity	ization	Toxicity	(log mM)	
	(mg/kg)			(log ug/L)		
Isoquinoline	2.189	No	Yes	0.148	0.972	
β -pinene	2.28	No	No	0.628	1.012	
Apigenin	2.298	No	No	0.38	2.432	
Salfredin	2.419	No	No	0.494	1.492	
B11	2.419	110	110	0.131	1.452	
Pyrazole	1.607	No	No	-1.123	3.048	
Pyragallol	2.374	No	No	0.127	2.734	
Salicylic	2.483	No	No	0.263	1.812	
Acid	2.405	NO	NO	0.205	1.012	
Syringic	2.157	No	No	0.281	2.554	
Acid	2.107	NO	NO	0.201	2.004	
Gallic	3.06	No	No	0.285	3.188	
Acid	5.00	110	INU	0.200	0.100	
Camphene	2.247	No	No	0.533	1.19	

	Oral Rat		Claim	Τ.	Minnow	
Ligands	Chronic	Hepato-	Skin Sensit-	pyriformis	Minnow toxicity	
Ligands	Toxicity	toxicity	ization	Toxicity	(log mM)	
	(mg/kg)			$(log \ ug/L)$	(log IIIvi)	
3,4-Dihydroxy-						
benzoic	2.021	No	No	0.273	2.451	
Acid						
4-Dihydroxy-						
cinnamic	2.534	No	No	0.319	1.607	
Acid						
Caffeic	2.092	No	No	0.293	2 246	
Acid	2.092	INO	INO	0.295	2.246	
Myristic	2 024	No	Yes	0.978	0.601	
Acid	3.034	INO	ies	0.978	-0.601	
Stearic	9 99	$\mathbf{N}_{\mathbf{c}}$	Yes	0.65	1 565	
Acid	3.33	No	res	0.65	-1.565	

TABLE 4.7: b). Toxicity predicted values of Ligands

4.5 Molecular Docking

Molecular docking is a technique that is used to estimate the strength of a bond between a ligand and a target protein through a special scoring function and also to find out the correct structure of the ligand within the target binding site. The 3D structure of the target proteins and the ligand is taken as input for docking. After preparing proteins and ligands ready for docking, docking is performed by CB-Dock which is well trusted online blind auto docking tool. The results and time required for docking is dependent upon structures of receptors, ligands, refinements and net speed. It may take several hours for a single result, so patience was shown while doing docking. CB dock gave us five possible possess and receptors models and among these possess best one was selected by observing certain properties like vina score and size of cavity etc.

Molecular docking without having information of binding sites is performed by using a user friendly blind docking web server called as CB Dock, that predicts and estimate a bonding site for a given protein and calculate centers and sizes with a novel rotation cavity detection method and perform docking with proper docking program known as Auto dock Vina [79]. Molecular dockings are performed by using ALK-EML4 as receptors and 15 selected compounds as ligands [80]. After submitting input files (receptor files in PDB format and ligand file in SDF format), CB-Dock checks the input files and convert them to PDB formatted files using Open Babel and MAGL Tools.

The molecular docking technique has recently become a crucial tool in computerassisted drug design to estimate the binding affinity and examine the interactive mode since it may significantly increase efficiency and lower research costs. Effective docking methods use a scoring system that correctly ranks candidate dockings and efficiently explore high-dimensional spaces. Lead optimization benefits greatly from the use of docking to do virtual screening on huge libraries of compounds, rate the outcomes, and offer structural ideas for how the ligands inhibit the target.

- After that CB-Dock predicts cavities of the receptor and calculated the centers and sizes of the top N (n=5 by default) cavities.
- Each center, size and PDB files are submitted by Auto Dock Vina for docking.
- Among 5 best confirmations, best one is selected on the basis of highest affinity score of ligand-receptor interaction.
- Ligands with best binding scores values with ALK and EML 4 receptors are shown in the table below 4.8.

Compounds	Binding Score	Cavity size	HBD	HBA	Log P	Molecular Weight (g/mol)	Rotatable Bond	Grid Map
Iso-quinoline	-5.8	1932	0	1	2.2348	129.16	0	27
β -pinene	-5.4	1932	0	0	2.9987	136.23	0	27
Apigenin	-7.9	1932	3	5	2.5768	270.24	1	27
Salfredin B11	-7.4	1932	1	4	2.2468	232.32	0	27
Pyrazole	-3.3	1932	1	1	0.4097	68.03	0	27
Pyragallol	-4.8	1932	3	3	0.8034	126.11	0	27
Salicylic Acid	-5.2	1932	2	2	1.0904	138.12	1	27

TABLE 4.8 :	Ligands	with be	st binding	score va	lues with ALK
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Compounds	Binding Score	Cavity size	HBD	HBA	Log P	Molecular Weight (g/mol)	Rotatable Bond	Grid Map
Syringic Acid	-5.5	1932	2	4	1.1076	198.17	3	27
Gallic Acid	-5.4	1932	4	4	0.5016	170.12	1	27
Camphene	-5.3	1932	0	0	2.997	136.23	0	27
3,4-Benzoic Acid	-5.2	1932	3	3	0.796	154.12	1	27
4-Dihydroxy Cinnamic Acid	-5.8	1932	2	2	1.49	164.16	2	27
Caffeic Acid	-6	1932	3	3	1.19	180.16	2	27
Myristic Acid	-5.2	1932	1	1	4.7721	228.37	12	27
Stearic Acid	-5.5	105	1	1	6.335	284.5	16	26

TABLE 4.8: Ligands with best binding score values with ALK

Compounds	Binding Score	Cavity Size	HBD	HBA	Log P	Molecular Weight (g/mol)	Rotatable Bond	Grid Map
Isoquinoline	-5.6	1415	0	1	2.2348	129.16	0	46
β -pinene	-5.5	11648	0	0	2.9987	136.23	0	66
Apigenin	-8.1	11648	3	5	2.5768	270.24	1	66
Salfredin B11	-7.2	11648	1	4	2.2468	232.32	0	66
Pyrazole	-4.2	11648	1	1	0.4097	68.03	0	66
Pyragallol	-5.7	11648	3	3	0.8034	126.11	0	66
Salicylic Acid	-5.7	11648	2	2	1.0904	138.12	1	66

TABLE 4.9: Ligands with best binding score values with EML4

Compounds	Binding Score	Cavity Size	HBD	HBA	Log P	Molecular Weight (g/mol)	Rotatable Bond	Grid Map
Syringic Acid	-6	11648	2	4	1.1076	198.17	3	66
Gallic Acid	-6.3	11648	4	4	0.5016	170.12	1	66
Camphene	-5.4	11648	0	0	2.997	136.23	0	66
3,4-Dihydroxybenzoic Acid	-6	11648	3	3	0.796	154.12	1	66
4-Dihydroxy Cinnamic Acid	-6.4	1415	2	2	1.49	164.16	2	46
Caffeic Acid	-6.5	11648	3	3	1.19	180.16	2	66
Myristic Acid	-5.6	11648	1	1	4.7721	228.37	12	66
Stearic Acid	-5.7	1415	1	1	6.335	284.5	16	46

TABLE 4.9: Ligands with best binding score values with EML4

4.6 Interaction of Ligands and Target Protein

The docking analysis is performed by using LigPlot+ (version v. 1.4.5) and PyMol Edu (v 1.7.4.5). Interactions of ligands and target proteins are predicted by using LigPlot plus (version v.1.4.5). The Graphical system of LigPlot+ automatically generates multiple 2D diagrams of interactions from 3D coordinates. These 2D diagrams portray the hydrogen-bond interaction pattern and hydrophobic contacts between the ligand and the main-chain or side-chain elements of the protein [81]. The 2D diagrams of the best binding score ligands with respective proteins are shown in Figures 4.7 to 4.12. while their hydrogen bonds and hydrophobic interactions are listed in Table 4.10 and 4.11.

4.6.1 Interaction of Ligands with Anaplastic Lymphoma Kinase

Figure Figures 4.7 shows the interaction of Isoquinoline, β -pinene, Apigenin, Salfredin B11 and Pyrazole with ALK. As observed from the 2D diagram ligand show only one hydrogen bond with Methionine. The ligand show many hydrophobic interactions with protein. The ligand contains 9 carbons and forms hydrophobic interaction with Glu1197, Leu1122, Val1130, Ala1148, Lys1150, Leu1196, Gly1269 residues. β -pinene show only hydrophobic interactions with protein. The ligand forms 9 hydrophobic interactions as shown in Figure 4.7. The ligand consist of 10 carbons and shows interaction with residues Glu1197, Leu1122, Val1130, Ala1148, Lys1150, Leu1196, Leu1256, Met1199 and Gly1269. Apigenin is a 15 carbons compound that forms 2 hydrogen bonds with the protein ALK. 7 residues form hydrophobic interactions with protein that includes Leu1196, Lys1150, Ala1148, Leu1198, Leu1122, Asp1203 and Leu1256. Figure 4.9 also shows the interaction of Salfredin B11 with protein ALK. One hydrogen bond is formed with Methionine. Salfredin B11 is a 13 carbons compound and forms 7 hydrophobic interactions with Ala1148, Leu1256, Leu1198, Lys1150, Val1130, Leu1122 and Gly1202. Pyrazole is a 3 carbons compound and forms 2 hydrogen bonds with Alanine and Glutamine. It forms 5 hydrophobic interactions with protein. These interactions are formed by residues Glu1197, Leu1198, Ala1148, Val1180, Leu1256 and Leu 1196. Figure 4.8 shows the interaction of Pyragallol, Salicylic acid, Syringic acid, Gallic acid, Camphene with ALK. Pyragallol is a 6 carbons sugar that forms one hydrogen bond with protein by methionine. 6 hydrophobic interactions are found in this interaction. The residues forms hydrophobic interactions are Glu1197, Leu1198, Ala1148, Val1180, Leu1256 and Leu1196. Salicylic acid is a 7 carbons compound and forms a single hydrogen bond with protein by Methionine. Many hydrophobic interactions are formed by Leu1198, Val1130, Leu1122, Ala1148, Leu1256 and Glu1197. Syringic acid 9 carbons compound form 1 hydrogen bond by methionine.

5 hydrophobic interactions are formed by residues includes Val1130, Lys1150, Ala1148, Leu1189, Leu112. Gallic acid is a 7 carbons compound and forms 2 hydrogen bonds by methionine and Glutamine. 5 residues involved in hydrophobic interactions are Leu1198, Ala1148, Val1130, Leu1256 and Lys1150. Camphene is a 10 carbons sugar and forms no hydrogen bond with protein. 7 hydrophobic interactions formed by camphene involved residues Val1130, Gly1269, Lys1150, Leu1196, Ala1148, Leu1122 and Leu1256. Figure 4.9 shows the interaction of 3,4-Dihydroxybenzoic acid, 4-Dihydroxycinnamic acid, Caffeic acid, Myristic acid and Stearic acid with ALK. 3,4-Dihydroxybenzoic acid is a 7 carbons compound and forms 1 hydrogen bond by methionine. 6 residues form hydrophobic interaction that includes Leu1198, Glu1197, Leu1256, Ala1148, Val1130 and Lys1150.

4-Dihydroxycinnamic acid is a 9 carbons compound that forms 1 hydrogen bond and 8 hydrophobic interactions. Hydrogen bond is formed by methionine and hydrophobic interactions are formed by residues Gly1123, Val1130, Leu1122, Leu1124, Ala1148, Leu1256, Leu1198 and Glu1197 as shown in Figure 4.9. Caffeic acid is a 9 carbons compound and forms 2 hydrogen bond and 8 hydrophobic interactions. Hydrogen bonds are formed by methionine and glutamine. Hydrophobic interactions are formed by residues Val1180, Leu1256, Ala1148, Val1130, Leu1122, Gly1123, Leu1198 and His1124. Myristic acid is a 14 carbons compound that forms a single hydrogen bond with protein and 11 hydrophobic interactions. Asn forms a hydrogen bond and residues involved in hydrophobic interactions are Arg1253, Leu1256, Gly1269, Asp1270, Ala1148, Glu1197, Leu1196, Met1199, Gly1269, Val1130 and Leu 1122.

Stearic acid is 18 carbons sugar and forms no hydrogen bond with protein but it is the ligands that form maximum hydrophobic interactions. 13 residues namely Lys1333, Pro1331, Tyr1330, Met1348, Asp1349, pro1350, Lys1352, Tyr1327, Trp1320, Ser1324, Pro1398, Ile1399 and Glu1400 forms hydrophobic interactions with protein ALK.

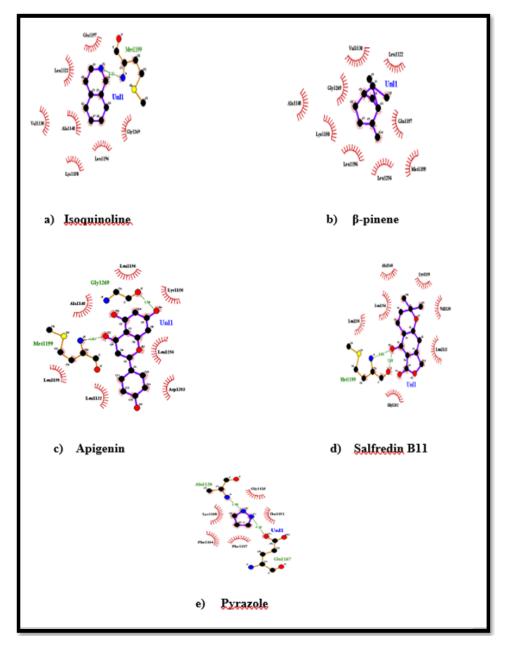


FIGURE 4.7: Interactions of ligands with the receptor protein ALK, a) Isoquinoline b) β -pinene, c) Apigenin, d) Salfredin B11, e) Pyrazole

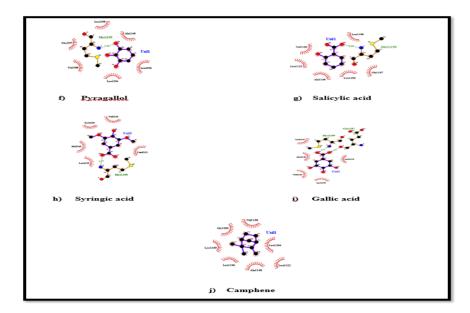


FIGURE 4.8: Interactions of ligands with the receptor protein ALK, f) Pyragallol, g) Salicylic acid, h) Syringic acid, i) Gallic acid, j) Camphene

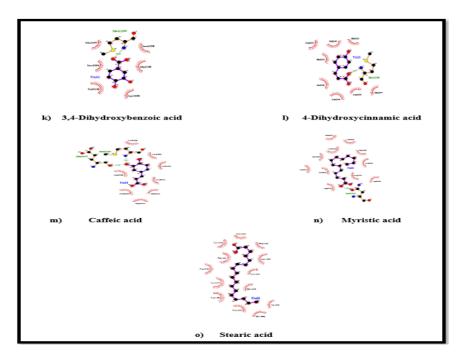


FIGURE 4.9: Interactions of ligands with the receptor protein ALK, k) 3,4-Dihydroxybenzoic acid, l) 4-Dihydroxycinnamic acid, m) Caffeic acid, n) Myristic acid

These 2D diagrams portray the hydrogen-bond interaction pattern and hydrophobic contacts between the ligand and the main-chain or side-chain elements of the protein [81]. The 2D diagrams of the best binding score ligands with respective proteins while their hydrogen bonds and hydrophobic interactions are listed in Table 4.10 and 4.11.

	Binding	No	Amino		Hydrophobic
Ligands		of	Acids	Distance	
	Energy	H.B	Acius		interaction
					Glu1197
					Leu1122
					Val1130
Isoquinoline	-5.8	1	Met1199	3.12	Ala1148
					Lys1150
					Leu1196
					Gly1269
					Glu1197
					Leu1122
	-5.4	0			Val1130
					Ala1148
β -pinene					Lys1150
					Leu1196
					Leu 1256
					Met1199
					Gly1269
			Met1199	2.96	Leu1196
Apigenin	-7.9	2	Gly1296	2.90 2.26	Lys1150
			GIy1290	2.20	Leu1122
					Ala1148
					Leu1256
Salfredin					Leu1198
B11	-7.4	1	Met1199	2.88	Lys1150
DII					Val1130
					Leu1122
					Gly1202

TABLE 4.10 :	Active Ligand Showing Hydrogen and Hydrophobic Interactions
	with ALK.

Binding	No	Amino		Hydrophobic
	of		Distance	
Energy	H.B		interaction	
				Gly1125
		Al-1196	2.00	Thr1151
-3.3	2			Lys1150
		GIUITO	3.21	Phe1164
				Phe1127
				Glu1197
				Leu1198
1 8	1	Mot1100	2 00	Ala1148
-4.0	1	Met1199	2.00	Val1180
				Leu1256
				Leu1196
				Leu1198
				Val1130
5.2	1	Mot1100	2.00	Leu1122
-0.2	T	1 1/1601133	2.33	Ala1148
				Leu1256
				Glu1197
				Val1130
-5.5	1	Met1199	2.97	Lys1150
				Ala1148
				Leu1198
		M_{a+1100}	914	Ala1148
-5.4	02			Val1130
		GIUI 197	2.31	Leu1256
				Lys1150
	Energy -3.3 -4.8 -5.2	of Energy H.B -3.3 2 -4.8 1 -5.2 1 -5.5 1	Amino of Benergy Amino Acids -anergy H.B -3.3 2 -4.8 1 -5.2 1 Met1199 -5.5 1 Met1199	Amino of AcidsDistanceEnergyH.BDistance-3.3 2 Ala1126 Glu11672.90 3.21-4.81Met11992.88-5.21Met11992.99-5.51Met11992.99-5.402Met11993.14

TABLE 4.10 :	Active Ligand	Showing Hydrogen	and Hydrophobic Interactions
		with ALK.	

	Binding	No	Amino		Hydrophobic
Ligands		of	Acids	Distance	
	Energy	H.B			interaction
					Val1130
					Gly1269
Camphene	-5.3	0			Lys1150
					Leu1196
					Ala1148
					Leu1198
					Glu1197
3,4-dihydroxy	-5.2	1	Met1199	3.12	Leu1256
benzoic acid	-0.2	1	Met 1199	3.12	Ala1148
					Val1130
					Lys1150
					Gly1123
					Val1130
					Leu1122
4-dihydroxy	-5.8	1	Met1199	2.87	Leu1124
cinnamic acid	-0.0	T	Met1199	2.01	Ala1148
					Leu1256
					Leu1198
					Glu1197
					Val1180
					Leu1256
Coffeir said	C	0	Met1199	2.83	Ala1148
Caffeic acid	-6	2	Glu1197	3.08	Val1130
					Leu1122
					Gly1123

TABLE 4.10 :	Active Ligand Sh	nowing Hydrogen	and Hydrophobic Interac	ctions
		with ALK.		

	Binding	No	Amino		Hydrophobic
Ligands		of	Acids	Distance	
	Energy	H.B	Acids		interaction
					Arg1253
					Leu1256
					Gly1269
					Asp1270
					Ala1148
Myristic acid	-5.2	1	Asn1254	2.83	Glu1197
					Leu1196
					Met1199
					Gly1269
					Val1130
					Leu1122
					Lys1333
					Pro1331
					Tyr1330
					Met1348
					Asp1349
					Pro1350
					Lys1352
Stearic acid	-5.5	0			Tyr1327
					Trp1320
					Ser1324
					Pro1398
					Ile1399
					Glu1400

TABLE 4.10 :	Active Ligand	Showing Hydrogen	and Hydrophobic Interactions
		with ALK.	

Ligands	Binding Energy	No of H.B	Amino Acids	Distance	Hydrophobic Interaction
					Glu1197
					Leu1122
Icocuin					Val1130
Isoquin- oline	-5.8	1	Met1199	3.12	Ala1148
onne					Lys1150
					Leu1196
					Gly1269
					Glu1197
					Leu1122
					Val1130
					Ala1148
β -pinene	-5.4	0	-	-	Lys1150
					Leu1196
					Leu1256
					Met1199
					Gly1269
					Leu1196
					Lys1150
					Ala1148
					Leu1198
A mi mari in	7.0	0	Met1199	2.96	
Apigenin	Apigenin -7.9	2	Gly1296	2.26	Leu1122
					Asp1203

TABLE 4.11: Active Ligand Showing Hydrogen and Hydrophobic Interactionswith ALK.

Ligands	Binding Energy	No of H.B	Amino Acids	Distance	Hydrophobic Interaction			
					Ala1148			
					Leu1256			
					Leu1198			
Salfredin	-7.4	1	Met1199	2.88	Lys1150			
B11					Val1130			
					Leu1122			
					Gly1202			
					Gly1125			
				2.00	Thr1151			
Pyrazole	-3.3	2	Ala1126	2.90	Lys1150			
			Glu1167	3.21	Phe1164			
					Phe1127			
					Glu1197			
				2.88	Leu1198			
	4.0		M. 1100		Ala1148			
Pyragallol	-4.8	1	Met1199		Val1180			
								Leu1256
					Leu1196			
					Leu1198			
					Val1130			
					Leu1122			
Salicylic	-5.2	1	Met1199	2.99	Ala1148			
Acid					Leu1256			
					Glu1197			
					Ala1148			

TABLE 4.11: Active Ligand Showing Hydrogen and Hydrophobic Interactionswith ALK.

Syringic Acid -5.5 1 Met1199 2.97 Val1130 Lys1150 Gallic Acid -5.4 2 Met1199 3.14 Glu1197 Leu1122 Leu1198 Gallic Acid -5.4 2 Met1199 3.14 Glu1197 Val1130 Leu1256 Camphene -5.3 0 - - Leu1196 Ala1148 Leu1256 Lys1150 Val1130 Gly1269 Leu1196 Ala1148 Leu1192 Leu1196 Ala1148 Leu1122 Leu196 Gallita - - Glu1197 3, Glu1197 - Glu1197	nobic ion
Syringic Acid -5.5 1 Met1199 2.97 Ala1148 Acid Leu1198 Leu1192 Leu1198 Gallic Acid -5.4 2 Met1199 3.14 Ala1148 Acid -5.4 2 Met1199 3.14 Ala1148 Acid -5.4 2 Met1199 3.14 Val1130 Glu1197 2.51 Leu1256 Lys1150 Camphene -5.3 0 - - Leu1198 Camphene -5.3 0 - - Leu1226 Leu1226 Leu1256 Leu196 Ala1148 Camphene -5.3 0 - - Heu1196 Ala1148 Leu1226 Leu1226 Leu1226 Leu1226 Leu1226 Leu1198 Leu1226 Leu1226 Leu1226 Leu1198 - - - Glu1197	
Acid -5.5 1 Met1199 2.97 Ala1148 Acid Leu1198 Gallic -5.4 2 Met1199 3.14 Acid -5.4 2 Met1199 3.14 Ala1148 Val1130 Glu1197 2.51 Leu1256 Lys1150 Val1130 Gly1269 Lys1150 Camphene -5.3 0 Leu1196 Ala1148 Leu1256 Lys1269 Lys150 Camphene -5.3 0 Glu1197 Camphene -5.3 0 Glu1197	
$ \begin{array}{c} {\rm Gallic} \\ {\rm Acid} & {}^{-5.4} & 2 \\ {\rm Camphene} & {}^{-5.3} & 0 \end{array} \begin{array}{c} {\rm Met1199} & {}^{-5.4} & 2 \\ {\rm Gu1197} & {}^{-5.1} & {}^{-5.4} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} & {}^{-1.5} &$	
Gallic Acid -5.4 2 Met1199 Gul1197 3.14 3.14 Ala1148 Acid -5.4 2 Glu1197 2.51 Leu1256 Lys1150 Val1130 Gly1269 Lys1150 Camphene -5.3 0 - - Leu1198 Kangelee -5.3 0 - - Leu1196 Gallita -5.3 0 - - Leu1196 Leu1196 - - - - Leu1196 Gallita - - - - - Gallita - - - - - Gallita - - - - - - - - - - - - - - - - - - - - - - - - - - - - - - - - - - - - <t< td=""><td></td></t<>	
Gallic Acid-5.42Met11993.14 Glu1197Ala1148 Val1130 Leu1256Acid-5.42Camphene-5.30Camphene-5.30Ala1148Ala1149Camphene-5.30Ala1148Ala1148Ala1148Ala1148Ala1148Ala1148Ala1148Ala1148Ala1148Ala1148Ala1148Ala1148Ala1148Ala1148Ala1148 <td></td>	
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	
Acid -5.4 2 Glu1197 2.51 Val1130 Leu1256 Lys1150 Val1130 Gly1269 Lys1150 Camphene -5.3 0 Leu1196 Ala1148 Leu1122 Leu1256 Leu1198	
Leu1256 Lys1150 Val1130 Gly1269 Lys1150 Camphene -5.3 0 Leu1196 Ala1148 Leu1122 Leu1226 Leu1256 Leu1198	
Val1130 Gly1269 Lys1150 Camphene -5.3 0 Leu1196 Ala1148 Leu1122 Leu1256 Leu1198	
Gly1269 Lys1150 Camphene -5.3 0 - Ala1148 Leu1122 Leu1256 Leu1198 Glu1197	
Camphene -5.3 0 Leu1196 Ala1148 Leu1122 Leu1256 Leu1198	
Camphene -5.3 0 Leu1196 Ala1148 Leu1122 Leu1256 Leu1198	
Ala1148 Leu1122 Leu1256 Leu1198 Glu1197	
Leu1122 Leu1256 Leu1198 Glu1197	
Leu1256 Leu1198 Glu1197	
Leu1198 Glu1197	
Glu1197	
3, Glu1197	
4-dihydroxy -5.2 1 Met1199 3.12 Leu1256	
benzoic Ala1148	
Acid Val1130	
Lys1150	
Lys1150 Leu1256	

TABLE 4.11: Active Ligand Showing Hydrogen and Hydrophobic Interactions with ALK.

Ligands	Binding Energy	No of H.B	Amino Acids	Distance	Hydrophobic Interaction
					Gly1123
					Val1130
1 dibuduoru					Leu1122
4-dihydroxy	-5.8	1	Met1199	2.87	Leu1124
	-0.0	T	Met1199	2.01	Ala1148
Acid					Leu1256
					Leu1198
					Glu1197
					Val1180
					Leu1256
C. C.	-6	2	Met1199 Glu1197	2.83 3.08	Ala1148
Caffeic					Val1130
Acid					Leu1122
					Gly1123
					Leu1198
					Arg1253
					Leu1256
					Gly1269
					Asp1270
M					Ala1148
Myristic	-5.2	1	Asn1254	2.83	Glu1197
Acid					Leu1196
					Met1199
					Gly1269
					Val1130
					Leu1122

TABLE 4.11: Active Ligand Showing Hydrogen and Hydrophobic Interactionswith ALK.

Ligands	Binding Energy	No of H.B	Amino Acids	Distance	Hydrophobic Interaction
					Lys1333
					Pro1331
					Tyr1330
					Met1348
				Asp1349	
C					Pro1350
Stearic	-5.5	0	-	-	Lys1352
Acid					Tyr1327
					Trp1320
					Ser1324
					Pro1398
					Ile1399
					Glu1400

TABLE 4.11: Active Ligand Showing Hydrogen and Hydrophobic Interactionswith ALK.

4.6.2 Interaction of Ligands with EML4

Figure 4.10 shows the interaction of Isoquinoline, β -pinene, Apigenin, Salfredin B11 and Pyrazole with EML4. As observed from the 2D diagrams the ligand show only hydrophobic interactions with protein. The ligand contains 9 carbons and forms hydrophobic interaction with Asn206, Tyr226, Asn228, Gln11, Gln15, Ala12, Ile16, Ile171 residues as evident from the table 4.12. β -pinene show only hydrophobic interactions with protein. The ligand forms 7 hydrophobic interactions. The ligand consist of 10 carbons and shows interaction with residues Ala330, Val177, Lys176, Phe214, Tys210, Lys326 and Asn329 as shown in Figure 4.10. Apigenin is a 15 carbons compound that forms 4 hydrogen bonds with the protein EML4 by glutamine, asparagine and glutamic acid. 4 residues form

hydrophobic interactions with protein that includes Tys224, Cys12, Leu248 and Glu11. Figure 4.10 also shows the interaction of Salfredin B11 with protein EML4. Two hydrogen bonds are formed with glutamic acid and asparagine. Salfredin B11 is a 13 carbons compound and forms 6 hydrophobic interactions with Gln11, Asp179, Leu248, Tyr224 Asn206 and Cys12. Pyrazole is a 3 carbon compound and forms 3 hydrogen bonds with leucine, asparagine and value. It forms 5 hydrophobic interactions with protein. These interactions are formed by residues Gly354, Ala240, Ser241, Asn356 and Ile355. Figure 4.11 shows the interaction of Pyragallol, Salicylic acid, Syringic acid, Gallic acid, Camphene with EML4. Pyragallol is a 6 carbons sugar that forms 4 hydrogen bonds with protein by glycine, asparagine, alanine and glycine. 6 hydrophobic interactions are found in this interaction. The residues forms hydrophobic interactions are Gly143, Asp69, Thr145, Glu71 and Glu254. Salicylic acid is a 7 carbons compound and forms 3 hydrogen bonds with protein by glutamine, glycine and threonine. 8 hydrophobic interactions are formed by Gly10, Gly144, Gly143, Glu254, Asn101, Glu71, Ala99 and Asp69. Syringic acid 9 carbons compound forms 4 hydrogen bonds by glycine, glutamine, threenine and asparagine. 5 hydrophobic interactions are formed by residues includes Gly144, Ser140, Gln11, Glu254 and Ala99. Gallic acid is a 7 carbons compound and forms 3 hydrogen bonds by asparagine, glycine and Glutamine. 7 residues involved in hydrophobic interactions are Gly143, Thr145, Asp69, Gly10, Glu71, Gly144 and Glu254.

Camphene is a 10 carbons sugar and forms no hydrogen bond with protein. 8 hydrophobic interactions formed by camphene involved residues Lys176, Val177, Lys326, Tyr20, Glu207, Asp211, Phe214 and Asn329. Figure 4.12 shows the interaction of 3,4-Dihydroxybenzoic acid, 4-Dihydroxycinnamic acid, Caffeic acid, Myristic acid and Stearic acid with EML4. 3,4-Dihydroxybenzoic acid is a 7 carbons compound and forms 4 hydrogen bonds by glycine, glutamine, threonine and asparagine.

6 residues form hydrophobic interaction that includes Gly144, Gly143, Gly10, Glu71, Glu254 and Ala99. 4-Dihydroxycinnamic acid is a 9 carbons compound that forms 1 hydrogen bond and 7 hydrophobic interactions. Hydrogen bond is

formed by tyrosine and hydrophobic interactions are formed by residues Gln14, Ala12, Ile16, Asn228, Ile171, Asn206 and Thr179. Caffeic acid is a 9 carbons compound and forms 5 hydrogen bond and 6 hydrophobic interactions. Hydrogen bonds are formed by alanine, glutamine, asparagine, selenocysteine and cysteine. Hydrophobic interactions are formed by residues Glu154, Gly144, Gly10, Glu11, Asp69 and Thr145. Myristic acid is a 14 carbons compound that forms a single hydrogen bond with protein and 12 hydrophobic interactions. Selenocysteine forms a hydrogen bond and residues involved in hydrophobic interactions are Lys326, Tyr210, Lys326, Lys176, Thr349, Phe351, Ile332, Ala333, Pro175, Val1177, Phe214 and Asn329. Stearic acid is 18 carbons sugar and forms 3 hydrogen bonds with protein by glycine, threonine and glycine. 12 residues namely Gln15, Gly10, Asn206, Thr126, Ile171, Gly143, Glu133, Tyr224, Asn228, Ser140, Thr179 and Asn228 forms hydrophobic interactions with protein EML4.The ligand contains 9 carbons and forms hydrophobic interaction with Asn206, Tyr226, Asn228, Gln11, Gln15, Ala12, Ile16, Ile171 residues as evident from the table 4.12.

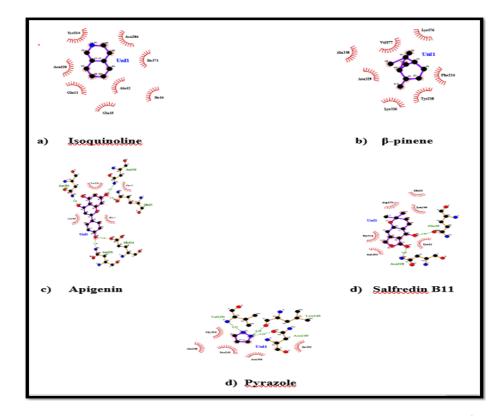


FIGURE 4.10: Interactions of ligands with the receptor protein EML4, a) Isoquinoline b) β -pinene, c) Apigenin, d) Salfredin B11, e) Pyrazole

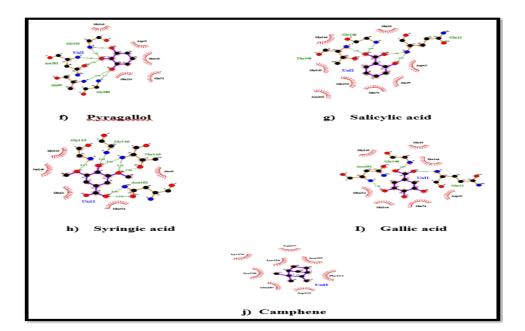


FIGURE 4.11: Interactions of ligands with the receptor protein EML4, f) Pyragallol, g) Salicylic acid, h) Syringic acid, i) Gallic acid, j) Camphene

Above interactions of ligands with the receptor protein EML4, f) Pyragallol, g) Salicylic acid, h) Syringic acid, i) Gallic acid, j) Camphene

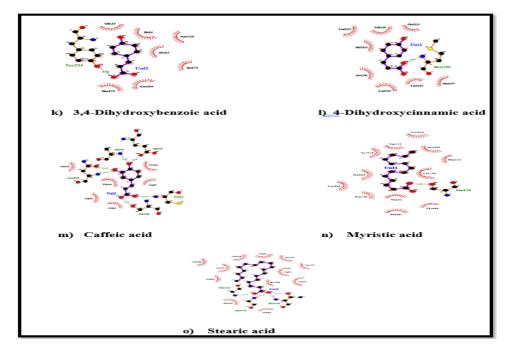


FIGURE 4.12: Interactions of ligands with the receptor protein ALK, k) 3,4-Dihydroxybenzoic acid, l) 4-Dihydroxycinnamic acid, m) Caffeic acid, n) Myristic acid

Above interactions of ligands with the receptor protein ALK, k) 3,4-Dihydroxybenzoic acid, l) 4-Dihydroxycinnamic acid, m) Caffeic acid, n) Myristic acid

Ligands	Binding Energy	No of H.B	Amino Acids	Distance	Hydrophobic Interaction
					Asn206
					Tyr226
					Asn228
Iso- quinoline	-5.6	0	-	-	Gln11
quinonne					Gln15
					Ala12
					Ile16
					Ile171
					Ala330
					Val177
					Lys176
β -pinene	-5.5	0	-	-	Phe214
					Tys210
					Lys326
					Asn329
					Tys224
			Asn226	2.15	
			Gln15	1.77	Cys12
Apigenin	-8.1	4	Asn206	1.51	
			Glu254	2.18	Leu248
			Asn101	2.10	
					Glu11

TABLE 4.12: Active Ligands Showing Hydrogen and Hydrophobic Interactions with EML4 $\,$

Ligands	Binding Energy	No of H.B	Amino Acids	Distance	Hydrophobic Interaction
Salfredin B11	-7.2	2	Gln15 Asn228	3.04 3.06	Gln11 Asp179 Leu248 Tyr224 Asn206 Cys12 Gly354
Pyrazole	-4.2	3	Leu248 Asn249 Val250	3.18 3.16 3.27	Ala240 Ser241 Asn356
Pyragallol	-5.7	4	Gly144 Asn102 Ala99 Gly100	3.01 2.30 3.06 3.13	Ile355 Gly143 Asp69 Thr145 Glu71 Glu254 Gly10
Salicylic Acid	-5.7	3	Gln11 Gly146 Thr145	3.03 1.96 1.99	Gly144 Gly143 Glu254

TABLE 4.12: Active Ligands Showing Hydrogen and Hydrophobic Interactions with EML4 $\,$

Asn101

Ligands	Binding Energy	No of H.B	Amino Acids	Distance	Hydrophobic Interaction
Syringic Acid	-6	4	Gly143 Gln146 Thr145 Asn101	2.88 3.17 2.88 3.01	Gly144 Ser140 Gln11 Glu254
Gallic Acid	-6.3	3	Asn101 Gly146 Gln11	3.09 1.50 1.95	Ala99 Gly143 Gly10 Thr145 Asp69 Glu71 Gly144 Glu254 Lys176
Camphene	-5.4	0	-	_	Val177 Lys326 Tyr20 Glu207 Asp211 Phe214 Asn329
3,4-dihydr- oxy benzoic Acid	-6	4	Gly146 Gln11 Thr145 Asn101	 2.78 2.91 3.16 3.05 	Gly144 Gly143 Gly10

TABLE 4.12: Active Ligands Showing Hydrogen and Hydrophobic Interactions with EML4

Ligands	Binding Energy	No of H.B	Amino Acids	Distance	Hydrophobic Interaction
4-dihydr- oxy cinnamic Acid	-6.4	1	Tyr224	3.01	Gln14 Ala12 Ile16 Asn228 Thr179
Caffeic Acid	-6.5	5	Ala99 Glu71 Asn101 Sec140 Cys12	3.04 2.94 2.03 3.02 3.03	Glu154 Gly144 Gly10 Glu11 Asp69 Thr145
Myristic Acid	-5.6	1	Sec178	2.51	Lys326 Tyr210 Lys326 Lys176 Thr349 Phe351 Ile332 Ala333
Stearic Acid	-5.7	3	Gly146 Thu145 Gly144	3.15 2.97 3.13	Gln15 Gly10 Asn206 Thr126 Ile171 Gly143 Glu133

TABLE 4.12: Active Ligands Showing Hydrogen and Hydrophobic Interactions with EML4 $\,$

4.7 ADME Properties of Ligands

Lipinski's five drug rule is the initial step in assessing the verbal bioavailability and artificial availability. A second study was performed by calculating the AD-MET properties of ligands as a measure of pharmacokinetics using the online tool pkCSM [82]. In pharmacology, there are two broad terms pharmacodynamics and pharmacokinetics which was discussed below.

4.7.1 Pharmacodynamics

Pharmacodynamics is the branch of pharmacology in which we study the effect of drugs on the body.

4.7.2 Pharmacokinetics

Pharmacokinetics is the branch of pharmacology in which we study the effect of body on drugs. We study the absorption of drugs, distribution of drugs, metabolism of drugs and excretion of drugs.

4.7.3 Absorption

In pharmacology especially pharmacokinetics, the transfer of drug from the blood stream into the tissues is called absorption. So, the chemical composition of a drug, as well as the environment in which drug is placed, work together to determine the rate and extend of drug absorption. Absorption is one of the ADME properties that determines the absorption of orally administered drugs and includes Water solubility, Intestinal absorption, Skin permeability, P-glycoprotein substrate and P- glycoprotein I and II inhibitors.

Water solubility (log S) of a compound predicts its solubility in water at 25C. It is predicted as a molar concentration logarithm (log mol/L). Lipid soluble drugs are less soluble in water than water soluble drugs.

The Caco-2 permeability model determines the logarithm of the apparent permeability coefficient (log Papp; log cm/s). A compound has a high Caco-2 absorbency if it has a Papp $>8 \ge 10-6 \text{ cm/s}$ (>0.9 in terms of pkCSM predictive value). Intestinal absorption predicts the percentage that will enter a person's small intestine. A compound with less than 30% absorption is less absorbent.

The Skin permeability depicts the absorbency in log Kp value, it has a valuable role in the formation of transdermal drugs. The element with the log Kp >-2.5 shows less skin penetration.

The P-glycoprotein substrates act as a natural barrier and removes toxins from the cell. This model predicts that the given compound is a P-glycoprotein (Pgp) substrate or not. If a compound show P-glycoprotein substrate then it may show low oral absorption. To reduce the absorption, P-glycoprotein can be easily removed from the cell.

P-glycoprotein I/II inhibitor model predicts that a compound may be a P-gp I/II inhibitor or not. P-gp inhibitors reduce the activity of P-gp and have high absorption.

Absorption properties as mentioned in the Table 4.13 and 4.14 shows that all the ligands show less water solubility. Caco2 permeability in the form of log Papp in 10-6 cm/S is within the normal range. The values of intestinal absorption values are good in the range. Skin permeability values in the form of log Kp are low. Apigenin, Salfredin B 11, Pyrazole and Syringic acid are predicted as P-glycoprotein substrate.

	Water	$CaCO_2$	Intestinal	Skin
Ligands	Solubility	Permeability	Absorption	Permeability
	$(\mathrm{mol/L})$	(cm/S)	(%)	$({ m Log}~{ m Kp})$
Isoquino-	-1.721	1.549	97.359	-1.824
line	-1.721	1.040	51.005	-1.024
β -pinene	-4.191	1.385	95.525	-1.653

TABLE 4.13: a)Absorption properties of ligands

	Water	CaCO ₂	Intestinal	Skin	
Ligands	Solubility	Permeability	Absorption	Permeability	
_	$(\mathrm{mol/L})$	(cm/S)	(%)	$({ m Log}~{ m Kp})$	
Apigenin	-3.329	1.007	93.25	-2.735	
Salfredin	-3.081	1.201	94.508	-3.236	
B11	0.001	1.201	54.000	0.200	
Pyrazole	0.178	1.579	90.415	-3.276	
Pyragallol	-1.408	1.122	83.549	-2.751	
Salicylic	-1.808	1.151	83.887	-2.723	
Acid	1.000	1.101	00.001	2.120	
Syringic	-2.223	0.495	73.076	-2.735	
Acid	2.220	0.450	10.010	2.100	
Gallic	-2.56	-0.081	43.374	-2.735	
Acid	2.00	0.001	10.01 1	2.100	
Camphene	-4.34	1.387	94.148	-1.435	
3,4-					
Dihydroxy-	-2.069	0.49	71.17	-2.727	
benzoic	2.000	0.10	11.11	2.1.2.1	
Acid					
4-					
Dihydroxy-	-2.378	1.21	93.49	-2.715	
cinnamic	2.010	1.21	00.10	2.110	
Acid					
Caffeic	-2.33	0.634	69.40	-2.722	
Acid	2.00	0.001	05.10	2.122	
Myristic	-4.952	1.56	92.691	-2.705	
Acid	1.004	1.00	02.001	2.100	
Stearic	-5.973	1.556	91.31	-2.726	
Acid	0.010	2.000	01.01		

TABLE 4.13: a) Absorption properties of ligands

	P-	Р-	Р-	
Ligands	Glycoprotein	Glycoprotein	Glycoprotein	
0	Substrate	Ι	II	
	Substrate		Inhibitor	
Isoquino-	No	No	No	
line	1.0	1.0		
β -pinene	No	No	No	
Apigenin	Yes	No	No	
Salfredin	No	No	No	
B11	110	110	110	
Pyrazole	Yes	No	No	
Pyragallol	No	No	No	
Salicylic	No	No	No	
Acid	110	110	110	
Syringic	Yes	No	No	
Acid	165	NO	NO	
Gallic	No	No	No	
Acid	NO	NO	NO	
Camphene	No	No	No	
3,4-				
Dihydroxy-	No	No	No	
benzoic	NO	NO	NO	
Acid				
4-				
Dihydroxy-	No	No	No	
cinnamic Acid				
Caffeic Acid	No	No	No	
Myristic Acid	No	No	No	
Stearic Acid	No	No	No	

 TABLE 4.14:
 b). Absorption properties of ligands

4.7.4 Distribution

Distribution in pharmacology is the branch of pharmacokinetics that deals with the movement of drugs in all over the body. Distribution as one of the ADME property includes four models namely as Volume of distribution in human (VDss expressed as log L/kg), Function unbound in humans (Fu), Blood brain barrier (BBB) permeability expressed as log PS [83].

	VDss	Fraction	BBB	CNS	
Ligands	-Human	Unbound	Permeability	Permeability	
	(L/Kg)	(Fu)	$(\log BB)$	$(\mathrm{Log}\;\mathrm{PS})$	
Isoquinoline	0.024	0.338	0.316	-1.91	
β -pinene	0.685	0.35	0.818	-1.857	
Apigenin	0.822	0.147	-0.734	-2.061	
Salfredin	0.363	0.465	-0.747	-2.827	
B11	0.505	0.405	-0.747	-2.021	
Pyrazole	-0.213	0.779	-0.242	-2.947	
Pyragallol	0.13	0.712	-0.441	-3.252	
Salicylic	-1.57	0.563	-0.334	-3.21	
Acid	-1.07	0.000	-0.004	0.21	
Syringic	-1.443	0.601	-0.191	-2.701	
Acid	-1.440	0.001	-0.191	-2.701	
Gallic	-1.855	0.617	-1.102	-3.74	
Acid	-1.655	0.017	-1.102	-3.74	
Camphene	0.547	0.354	0.787	-1.71	
3,4-Dihydroxy-					
benzoic	-1.298	0.648	-0.683	-3.305	
Acid					
4-Dihydroxy-					
cinnamic	-1.151	0.428	-0.225	-2.418	
Acid					

TABLE 4.15: Distributive properties of ligands

	VDss	Fraction	BBB	CNS	
Ligands	-Human	Unbound	Permeability	Permeability	
	(L/Kg)	(Fu)	$(\log BB)$	$(\mathrm{Log}\;\mathrm{PS})$	
Caffeic	-1.098	0.529	-0.647	-2.608	
Acid	-1.030	0.029	-0.047	-2.000	
Myristic	-0.578	0.171	-0.027	-1.925	
Acid	-0.078	0.171	-0.027	-1.920	
Stearic	-0.528	0.051	-0.195	-1.707	
Acid	-0.020	0.001	-0.130	-1.707	

TABLE 4.15: Distributive properties of ligands

Model-I explains the theoretical volume that the total amount of drug will need to be evenly distributed to provide the same concentration in blood plasma. VDss is considered low, if it is less than 0.71 L/kg (log VDss <0.15) and higher if it is above 2.81 L/kg (log VDss >0.45). If VDss is high, it means that more drug dispense to the tissues rather to plasma. If a compound shows more Fu value, its mean it is more effective. BBB protects the brain from exogenous compounds, so BBB permeability is an important parameter 4.15.

4.7.5 Metabolism

CYP1A2, CYP2C19, CYP2C9, CYP2D6 and CYP3A4 models of the various isoforms of Cytochrome P450 that act as an important cleansing enzyme found in the liver. This enzyme reacts to xenobiotic to facilitate their release. Some drugs are triggered by this enzyme while most drugs are neutralized by it.

The ligands listed in table 4.16 do not act as substrate of any isoform except Stearic acid act as CYP2D6 substrate while Pyragallol and Stearic acid act as CYP3A4 substrate. The ligands Isoquinoline and Apigenin act as inhibitor of CYP1A2 isoform. Stearic acid acts as an inhibitor of CYP3A2 isoform.

TABLE 4.16: Metabolic	properties	of ligands
--------------------------	------------	------------

	CYP2D6	CYP3A4	CYP1A2	CYP2C19	CYP2C9	CYP2D6	CYP3A4
Ligands							

	Substrate	Substrate	Inhibitor	Inhibitor	inhibitor	Inhibitor	Inhibitor
Isoquinoline	No	No	Yes	No	No	No	No
β -pinene	No						
Apigenin	No	No	Yes	No	No	No	No
Salfredin	No						
B11	NO						
Pyrazole	No						
Pyragallol	No	Yes	No	No	No	No	No
Salicylic	No						
Acid	NO	INO	NO	NO	NO	INO	NO
Syringic	No						
Acid	NO						
Gallic	No						
Acid	INO	INU	NO	110	INU	no	INU
Camphene	No						

TABLE 4.16 :	Metabolic	properties	of ligands
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CYP2D6 CYP3A4 CYP1A2 CYP2C19 CYP2C9 CYP2D6 CYP3A4 Ligands

	Substrate	Substrate	Inhibitor	Inhibitor	inhibitor	Inhibitor	Inhibitor
3,4-Dihydroxy-							
benzoic	No						
acid							
4-Dihydroxy-							
cinnamic	No						
Acid							
Caffeic	No	No	No	No	Ne	No	Ne
Acid	INO	NO	No	INO	No	INO	No
Myristic	No						
Acid	INO	No No	INO	INO	NO	INO	INO
Stearic	Yes	Yes	No	No	No	No	No
Acid	res	res	INO	INO	INO	INO	No

4.7.6 Excretion

The organs involved in drug excretion are the kidneys, which play important role in excretion (renal excretion) and the liver (biliary excretion). Other organs may be also be involved in excretion, such as the lungs for volatile or gaseous agents. Drugs can be secreted in sweat, saliva and tears. Models of Excretion property are Total Clearance (CL tot) expressed as log (CL tot) in ml/min/kg and second one is Renal OCT2 substrate which predicts results as Yes/No. OCT2 (organic cation transporter 2) is a renal uptake transporter that plays role in disposition and renal clearance of drugs [85].

All ligands exhibit well total clearance. All ligands showed negative result for model Renal OCT2 substrate except. Excretory properties are listed in Table 4.17

Ligands	Total Clearance	Renal OCT2
	$(\mathrm{ml/kg})$	Substrate
Isoquinoline	0.286	No
β -pinene	0.03	No
Apigenin	0.566	No
Salfredin B11	0.481	No
Pyrazole	0.62	No
Pyragallol	0.104	No
Salicylic acid	0.607	No
Syringic acid	0.646	No
Gallic acid	0.518	No
Camphene	0.049	No
3,4-Di hydroxy benzoic acid	0.551	No
4-Dihydroxy cinnamic acid	0.662	No
Caffeic acid	0.508	No

TABLE 4.17: Excretory properties of ligands

Ligands	Total Clearance	Renal OCT2
	(ml/kg)	Substrate
Myristic acid	1.693	No
Stearic acid	1.832	No

	TABLE 4.17 :	Excretory	properties	of ligands
--	----------------	-----------	------------	------------

4.8 Lead Compound Identification

The identification of compound as a drug or non-drug is determined by Physiochemical and Pharmacokinetics properties. Physiochemical properties or Lipinski's rule act as first or a primary filter and then pharmacokinetics comes that sorts further potential compounds. Gallic acid do not obey Lipinski's rule of five, so it is knock out in primary screening. Pharmacokinetic studies screen out the compounds Isoquinoline, β -pinene and Camphene (log BB >0.3).

Name	Binding
of	Score
Potential	\mathbf{with}
Compound	ALK
Apigenin	-7.9
Salfredin B11	-7.4
Syringic Acid	-5.5
4-Dihydroxy- benzoic Acid	-5.8
Caffeic Acid	-6

TABLE 4.18: Hit compounds with binding scores with ALK.

Name	Binding
of	Score
Potential	with
Compound	ALK
Apigenin	-8.1
Salfredin B11	-7.2
Syringic Acid	-6
4-Dihydroxy-cinnamic Acid	-6.4
Caffeic Acid	-6.5

TABLE 4.19: Hit compounds with binding scores with EML4.

Myristic and stearic acid is screen out due to skin sensitivity property. The best compounds are selected on the basis of primary and secondary filters, toxicity predicted values and binding scores. These are Apigenin, Pyrazole, Pyragallol, Salicylic acid, Syringic acid and Caffeic acid.

4.9 Selection of Lead Drugs

Selection of the most efficient drug is based on the physiochemical properties that include molecular formula, molecular weight, water solubility, log P value, absorption, hydrogen bond donors and acceptors, polarization, bioavailability and ADMET probability. The side effects of the selected drugs were also studied using Drug bank database, PubChem and pkCSM tools. Mechanisms of action of selected drugs are listed in Table 4.43. Crizotinib has been selected as a standard drug against ALK receptors which is known very commonly in treating Lungs cancer [86]. Paclitaxel has been selected as standard drug against EML4. It is a standard drug used to treat lung cancers. It acts with the novel mechanism of actions by promoting polymerization of tubulin dimers to form microtubules and stabilizes microtubules by preventing depolymerisation [87].

4.10 Reference Drugs Actions:

4.10.1 Crizotinib Action against ALK

Crizotinib is a tyrosine kinase (TK) inhibitor and target ALK under its activation. ALK inhibits the apoptosis and induces cell proliferation. ALK gene translocation leads to the expression of proteins involved in cancer/oncogenic fusion. In most cases of NSCLC, ALK fusion with EML4 results in kinase activity. Crizotinib inhibits the activity of ALK by inhibiting its phosphorylation, results in inactive protein confirmation. This reduces the growth of cells having this genetic mutation and tumor survivability [88].

4.10.2 Paclitaxel action against EML4

Paclitaxel had reported to induce ROS (Reactive Oxygen Species) generation and it increase hydro peroxide production by increasing the NADPH (Nicotinamide Adenine Dinucleotide Phosphate) oxidase activity that induces oxidative stress and also play a role in anticancer activity. Paclitaxel targets the microtubule proteins. Paclitaxel high concentration causes mitotic arrest at G2 or M phase, while its low concentration induces apoptosis at G0 and G1or S phase. Paclitaxel induce proapoptotic activity by activating multiple signaling pathways [89].

4.11 Physiochemical Properties of Drugs

Physiochemical properties of the selected drugs Crizotinib and Paclitaxel for ALK and EML4 respectively are shown in 4.20

Properties	Crizotinib	Paclitaxel
Chemical	C ₂₁ H ₂₂ Cl ₁₂ FN ₅ O	C H NO-
Formula	$O_{21} II_{22} OI_{12} II_{5} O$	$O_{47}I151I1O_{14}$

TABLE 4.20: Physiochemical properties of Drugs.

Properties	Crizotinib	Paclitaxel
Absorption	5 hours	7 hours
Water Solubility	$0.5 \mathrm{~mg/ml}$	0.0056 mg/ml
mg/ml log P H-bond	5.0377 02	3.7357 4
Donor H-bond Acceptor	06	14
Molecular Weight	450.345 g/mol	853.9 g/mol
Rotatable Bonds	5	10
Bioavailability	1	0
Polarizability	45.04 Å3	87.15 Å3
Side	Sensitivity to light, numbness, difficulty falling asleep, dark urino	Redness of face, neck, arm, occasionally upper neck,
Effects	urine, nausea, vomiting and difficulty in swallowing	bruising, unusual bleeding tiredness and blurred vision.

TABLE 4.20: Physiochemical properties of Drugs.

4.12 2D Structure of Reference Drugs

2D structures of both the reference drugs (Crizotinib for ALK and Paclitaxel for EML4) were obtained from the online tool PubChem.

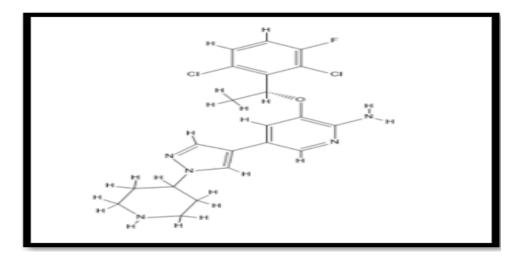


FIGURE 4.13: 2D Structure of Crizotinib Drug- PubChem

Figure 4.13 shows the bonding pattern of Drug Crizotinib for ALK protein. The simplified 2D structure of drug is used to display molecular configuration, profiles and interactions.

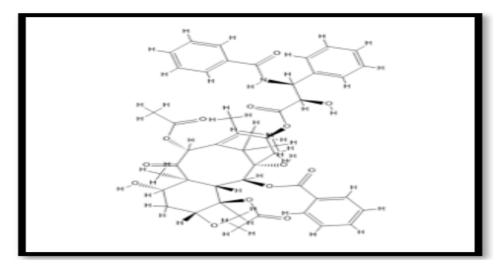


FIGURE 4.14: 2D Structure of Crizotinib Drug- PubChem

Figure 4.14 shows the bonding pattern of Drug Paclitaxel for EML4 protein. The simplified 2D structure of drug is used to display molecular configuration, profiles and interactions.

4.13 Drug ADMET Properties

Online tool pkCSM determines the ADMET properties (Absorption, Distribution, Metabolism, Excretion and Toxicity) of the reference drugs (Crizotinib and Paclitaxel).

4.13.1 Toxicity Prediction of Reference Drug-Crizotinib

The toxicity predictions of reference drug Crizotinib are listed in Table 4.21. The maximum tolerated dose value is shown as -0.095. The drug predicts as hERG II inhibitor which determines that it inhibits potassium channels. LD50 determines the potential of drug and LOAEL predicts the lowest dose that causes adverse effects. Crizotinib is a hepatotoxic that's mean it does cause liver injury.

T.pyriformis toxicity measures toxic end point (pIGC50 that is negative logarithm of the concentration required to stop 50% growth >-0.5 is considered as toxic). Crizotinib predicts pIGC out of this range.

The last model Minnow toxicity predicts LC50 in mM that shows the lethal concentration of a molecule that is enough to kill 50% flathead minnows (flat bait fishes). Crizotinib predicts minnow toxicity value as 0.942 log mM.

4.13.2 Absorption Properties

Absorption properties of Crizotinib are shown in Table 4.21. As evident from the table,

• Crizotinib is less soluble in water and has 92.006% absorption in small intestine of human. Skin permeability is low. Crizotinib is a P-gp substrate and P-gp I/II inhibitor. It means that standard drug has low Oral absorption.

P-gp I/II inhibitor 'Yes' predicts that Crizotinib has reduced pumping activity to pump out the xenobiotic from cell and have high absorption.

4.13.3 Distribution Properties

Distribution properties consist of four models. The first is the volume of distribution in human (VDss) expressed as L/Kg. Crizotinib Shows high VDss =0.801 L/Kg that means the drug is more distributed in tissue rather plasma. Second model is Fraction unbound (Fu) determines the unbound friction in plasma.

It is more than drug may be more effective. Crizotinib has 0.132 Fu predicted value. The third model is BBB permeability (blood brain barrier permeability) is expressed as log BB (value >-1 predicts that drug is not safe for brain). Crizotinib shows BBB permeability -1.164 slightly higher than -1. The last model is CNS permeability (Central Nervous System permeability) is expressed as log PS <-3 considered as safe. Crizotinib shows log PS -1.473 (Table 4.21).

4.13.4 Metabolic Properties

Reference drug Crizotinib metabolic properties are given in 4.21. Cytochrome P450 is found in liver that has a detoxification function and plays a role in excretion of exogenous compounds by oxidizing them.

CYP2D6 and CYP3A4 are the two main isoforms of cytochrome P450. Crizotinib is metabolized by one isoform CYP3A4. This drug is also an inhibitor of CYP1A2, CYP2C9 and CYP3A4. Crizotinib is not an inhibitor of CYP2C19 and CYP2D6 isoform of cytochrome P450.

4.13.5 Excretion Properties

The predicted value of excretion properties of Crizotinib are given in 4.21. Total clearance is expressed as log (CL tot) value is 0.556 ml/kg that predicts the hepatic and renal clearance of Crizotinib. OCT2 is an organic cation transporter 2 that play a role in disposition and renal clearance of drugs. Crizotinib shows Renal OCT2 substrate 'Yes' which means it interfere the function of OCT2 in the cell.

ADMET	Model	Predicted
Properties	Name	Values
	AMES Toxicity Max.	No
	Tolerated	-0.095
Toxicity	Dose (human)	m mg/Kg
	hERG I Inhibitor	No
	hERG II Inhibitor	Yes
	Oral rat acute Toxicity	3.515 mol/Kg
	Oral rat chronic toxicity	1.57 mg/kg
	Hepatotoxicity	Yes
	Skin sensitization	No
	T.pyriformis	0.296
	toxicity	$\log\mathrm{ug/L}$
	Minnow	0.942
	toxicity	$\log \mathrm{mM}$
	Water	-4.14
	solubility	$\mathrm{mol/L}$
	Caco2	0.702
Absorption	permeability	m cm/S

 TABLE 4.21: ADMET properties of reference drug- Crizotinib.

ADMET	Model	Predicted
Properties	Name	Values
	Intestinal absorption (human)	92.006%
	Skin permeability	-2.747 log Kp
	P-glycoprotein substrate	Yes
	P-glycoprotein I	Yes
	inhibitor P-glycoprotein	
	II inhibitor	Yes
	VDss	0.801
	(human)	L/Kg
Distribution	Fraction unbound	0.132
	(human)	Fu
	BBB	-1.164
	permeability	$\log BB$
	CNS	-2.222
	permeability	Log PS
	CYP2D6	No
	substrate CYP3A4	Yes
Metabolism	substrate CYP1A2	
		Yes

TABLE 4.21: ADMET properties of reference drug- Crizotinib.

inhibitor

ADMET	Model	Predicted
Properties	Name	Values
	CYP2C19	No
	inhibitor	NO
	CYP2C9	Yes
	inhibitor	Tes
	CYP2D6	No
	inhibitor	No
	CYP3A4	Yes
	inhibitor	105
	Total	
Excretion	Clearance	0.571
	(ml/kg)	
	Renal	
	OCT2	Yes
	substrate	

TABLE 4.21: ADMET properties of reference drug- Crizotinib.

4.13.6 Toxicity Prediction of Reference Drug- Paclitaxel

The toxicity prediction of reference drug Paclitaxel is listed in 4.22. The maximum tolerated dose value is shown as 0.199. The drug predicts as hERG II inhibitor which determines that it inhibit potassium channels. LD50 determines the potential of drug and LOAEL predicts the lowest dose that causes adverse effects.

Paclitaxel is non-hepatotoxic that's mean it does not cause liver injury. *T.pyriformis* toxicity measures toxic end point (pIGC50 that is negative logarithm of the concentration required to stop 50% growth >-0.5 is considered as toxic).

Paclitaxel predicts pIGC out of this range. The last model Minnow toxicity predicts LC50 in mM that shows the lethal concentration of a molecule that is enough to cause death of 50% flathead minnows (flat bait fishes). Paclitaxel predicts minnow toxicity value as 2.988 log mM.

4.13.7 Absorption Properties

Paclitaxel shows absorption properties as shown in 4.22. As shown in table, Paclitaxel is less soluble in water and has 100% absorption in small intestine of human. Skin permeability is low. Paclitaxel is a P-gp substrate and P-gp I/II inhibitor. It means that standard drug has low Oral absorption. P-gp I/II inhibitor 'Yes' predicts that Paclitaxel has reduced pumping activity to pump out the xenobiotic from cell and have high absorption.

4.13.8 Distribution Properties

Distribution properties consist of four models. The first is the volume of distribution in human (VDss) expressed as L/Kg. Paclitaxel Shows high VDss =1.458 L/Kg that means the drug is more distributed in tissue rather plasma. Second model is Fraction unbound (Fu) determines the unbound friction in plasma. It is more than drug may be more effective. Paclitaxel has 0.132 Fu predicted value. The third model is BBB permeability (blood brain barrier permeability) is expressed as log BB (value >-1 predicts that drug is not safe for brain). Paclitaxel show BBB permeability as -1.164. The last model is CNS permeability (Central Nervous System permeability) is expressed as log PS <-3 considered as safe. Paclitaxel shows log PS -1.473. The distribution properties are listed in 4.22.

4.13.9 Metabolic Properties

The Reference drug Paclitaxel metabolic properties are given in 4.22. Cytochrome P450 is found in liver that has a detoxification function and plays a role in excretion of exogenous compounds by oxidizing them. CYP2D6 and CYP3A4 are the two main isoforms of cytochrome P450. Paclitaxel is metabolized by one isoform

CYP3A4. This drug is also an inhibitor of CYP3A4. Paclitaxel is not an inhibitor of CYP2C19 and CYP2D6 isoform of cytochrome P450.

4.13.10 Excretion Properties

The predicted value of excretion properties of Paclitaxel are given in 4.22. Total clearance is expressed as log (CL tot) value is -0.36 ml/kg that predicts the hepatic and renal clearance of Paclitaxel. OCT2 is an organic cation transporter 2 that play a role in disposition and renal clearance of drugs. Paclitaxel shows Renal OCT2 substrate 'No' which means it is interfering in the function of OCT2 in the cell.

ADMET	Model	Paclitaxel
Properties	Name	I aciitaxci
	AMES	No
	toxicity	110
	Max.	0 100
	tolerated dose	0.199
	(human)	m mg/Kg
Toxicity	hERG I	No
	inhibitor	NO
	hERG II	Yes
	inhibitor	Tes
	Oral	0.776
	rat acute	2.776
	toxicity	m mol/Kg
	Oral	
	rat	3.393
	chronic	mg/kg
	toxicity	
	Hepatotoxicity	Yes

TABLE 4.22: ADMET properties of reference drug-Paclitaxel.

		0
ADMET	Model	Paclitaxel
Properties	Name	I actitaxei
	Skin	No
	sensitization	NO
	T.pyriformis	0.285
	toxicity	$\log \mathrm{ug/L}$
	Minnow	2.988
	toxicity	$\log \mathrm{mM}$
		-3.158
	Water solubility	$\mathrm{mol/L}$
Absorption	Caco2	0.623
	permeability	$\mathrm{cm/S}$
	Intestinal	
	absorption	100%
	(human)	
	Skin	-2.735
	permeability	log Kp
	P-glycoprotein	Yes
	substrate	res
	P-glycoprotein	Vee
	I inhibitor	Yes
	P-glycoprotein	
	II	Yes
	inhibitor	
	VDss	1.458
	(human)	L/Kg
Distribution	Fraction	
	unbound	0 Fu
	(human)	

TABLE 4.22: ADMET properties of reference drug-Paclitaxel.

ADMET	Model	Paclitaxel	
Properties	Name	I aciitaxei	
	BBB	-1.731	
	permeability	$\log BB$	
	CNS	-3.95	
	permeability	Log PS	
	CYP2D6		
	substrate	No	
	CYP3A4	37	
	substrate	Yes	
Metabolism	CYP1A2	.	
	inhibitor	Yes	
	CYP2C19	N	
	inhibitor	No	
	CYP2C9	Yes	
	inhibitor	Tes	
	CYP2D6	No	
	inhibitor	NO	
	CYP3A4	Yes	
	inhibitor	105	
	Total		
Enoutin	Clearance	-0.36	
Excretion	(ml/kg)		
	Renal		
	OCT2	No	
	substrate		

 TABLE 4.22: ADMET properties of reference drug-Paclitaxel.

4.14 Mechanism of Actions of Standard Drugs

4.14.1 Crizotinib Mechanism of Action

Figure 4.15 shows the mechanism of action of Crizotinib .Crizotinib is a tyrosine kinase (TK) inhibitor and target ALK under its activation. ALK inhibits the apoptosis and induces cell proliferation. ALK gene translocation leads to the expression of proteins involved in cancer/oncogenic fusion. In most cases of NSCLC, ALK fusion with EML4 results in kinase activity. Crizotinib inhibits the activity of ALK by inhibiting its phosphorylation, results in inactive protein confirmation. This reduces the growth of cells having this genetic mutation and tumor survivability [90]. Crizotinib is an orally bioavailable molecule that resist the growth of tumors with ALK activity. Crizotinib induces down regulation of STAT3 phosphorylation along with the significant apoptotic death. Apoptosis induces by the caspase-3 cleavage and down regulation of Bcl-2 family of proteins. Thus Crizotinib has the potential to treat patients with ALK induce Lung cancer [91].

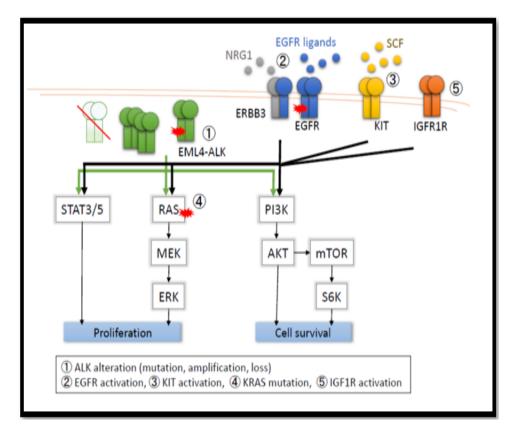


FIGURE 4.15: Mechanism of action of Crozotinib [92].

4.14.2 Paclitaxel Mechanism of Action

Figure 4.16 shows the mechanism of action of Paclitaxel. Paclitaxel had reported to induce ROS (Reactive Oxygen Species) generation and it increase hydro peroxide production by increasing the NADPH (Nicotinamide Adenine Dinucleotide Phosphate) oxidase activity that induces oxidative stress and also play a role in anticancer activity. Paclitaxel targets the microtubule proteins. Paclitaxel high concentration causes mitotic arrest at G2 or M phase, while its low concentration induces apoptosis at G0 and G1 or S phase [93]. Paclitaxel induce proapoptotic activity by activating multiple signaling pathways. In vitro, Paclitaxel increases the polymerization of tubulin to stable microtubule. The drug 'Paclitaxel' has a specific binding site on microtubule polymer makes it a different and uniform chemotherapeutic agent. Paclitaxel does not have the ability to polymerize tubulin in the absence of cofactors (like Guanosine triphosphate and microtubule associated protein). Paclitaxel and microtubule proteins are irradiated with UV light and the drug binds with the beta-subunit of tubulin. The microtubule pathway reorganizes in the presence of Paclitaxel [94].

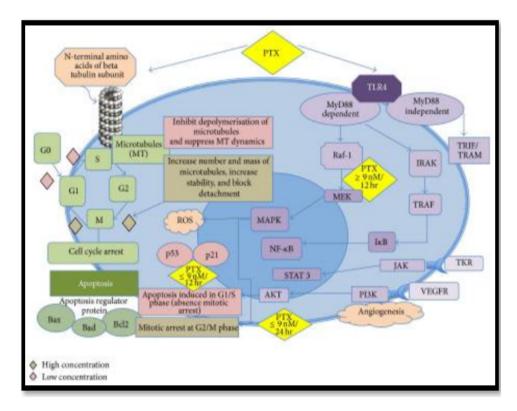


FIGURE 4.16: Mechanism of action of Paclitaxel [95].

4.15 Effects of Standard Drugs on Body

4.15.1 Crizotinib Effects on Body

Crizotinib is a novel tyrosine kinase inhibitor approved globally for treating patients with advanced or metastatic Non-Small Cell Lung Cancer. Side effects include acne like rashes, dryness, discoloration, perifollicular inflammation, acral erythema, alopecia, visual impairment, peripheral edema etc. Photo allergic dermatitis is less common with Crizotinib [96].

4.15.2 Paclitaxel Effects on Body

Paclitaxel has been used to treat lung cancer. Side effects of the drugs includes anemia, neutroprenia and alopecia. Patients taking Pclitaxel may report indigestion, viral infection, weakness ,nausea, vomiting and diarrhea [97].

4.16 Docking Results of Standard Drugs

4.16.1 Crizotinib Docking

Docking was performed with Crizotinib and paclitaxel as ligands by online docking tool (CB dock). Drug target was ALK and EML4 receptors respectively. Best docking score was -8.5 with ALK and -8.2 with EML4. Molecular interactions of docked drugs with targets are listed below in 4.23.

Docking Score	Crizotinib	Paclitaxel
	\mathbf{with}	\mathbf{with}
	ALK	EML4
Binding	-8.5	-8.2
scores		

TABLE 4.23: Docking Score of reference drugs via CB Dock.

Dealling	Crizotinib	Paclitaxel
Docking	with	\mathbf{with}
Score	ALK	EML4
Cavity	1932	11648
size	1552	11040
HBD	2	14
HBA	6	4
Log	5 0277	2 7257
Р	5.0377	3.7357
Molecular		
weight	450.345	853.918
(g/mol)		
Rotatable	5	10
bonds	0	10
Grid	00	66
map	23	66
Min energy	4.65	17 19
(kcal/mol)	4.00	17.13
Max energy(kcal/mol)	41.67	168.08

TABLE 4.23: Docking Score of reference drugs via CB Dock.

4.17 Standard Drugs and Lead Compounds Comparison

The Standard drug and lead compounds were compared for their physiochemical and pharmacokinetic properties to assess their drug likeness, bioavailability, efficacy and their safety. All these compounds passed the drug likeness criteria. Apigenin has low molecular weight and log P value and is a 3 HBD and 5 HBA whereas Crizotinib shows that it is a 2 HBD and 6 HBA (4.24).

Name of	Log P	Molecular	H-Bond	H-Bond
Compound	value	Weight	Donor	Acceptor
		270.24		
Apigenin	2.5768		3	5
		g/mol		
		450.345		
Crizotinib	5.0377		2	6
		g/mol		

TABLE 4.24: Apigenin- Crizotinib Lipinski's Rule of Five

Salfredin B11 has low molecular weight and log P value and is a 1 HBD and 4 HBA whereas Paclitaxel shows that it is a 4 HBD and 14 HBA (4.25).

Name of	Log P	Molecular	H-Bond	H-Bond
Compound	Value	Weight	Donor	Acceptor
Salfredin	2 2469	232.235	1	4
B11	2.2468	g/mol	1	4
	0 7057	853.918	4	14
Paclitaxel	3.7357	g/mol	4	14

TABLE 4.25: Salfredin B11- Paclitaxel Lipinski's Rule of Five

4.17.1 ADMET Properties Comparison of Crizotinib and Apigenin

Pharmacokinetics properties include Absorption, Distribution, Metabolism, Excretion and Toxicity (ADMET) properties play an important role I screening of compounds as drug candidates. Pharmacokinetic properties of reference drugs and lead compound are listed in Table 4.26 and 4.27.

Toxicity is the important parameter of pharmacokinetics (ADMET) properties which consist of 10 models. Model 1 of AMES toxicity depicts the standard drug and lead compound are not mutagenic. Maximum tolerated dose helps to set maximum recommended tolerated dose if value is $\leq 0.477 \log \text{mg/kg/day}$ then considered low and greater values are considered high. Table 4.23 shows -0.095mg.kg value for Crizotinib and 0.328mg/kg for Apigenin that depicts the bioactive compound Apigenin is safe to use. The model hERG I/II inhibitor depicts that the compounds are inhibitor of potassium channel or not. Crizotinib show itself as hERG II inhibitor. The model Oral rat acute toxicity (LD50) expressed as mol.kg is the amount of drug that cause the death of 50% rats. LD50 value of Crizotinib is higher than Apigenin. Oral rat chronic toxicity (LOEAL) determines the lowest dose of drug which can produce adverse effects of drug over long duration. LOEAL predicted value of Crizotinib is less than Apigenin which shows its potency to be less toxic than bio compound. Hepatotoxicity indicates the injury to liver. Crizotinib shows that it is hepatotoxic while Apigenin is non-hepatotoxic. Both compounds do not cause any allergic reactions.

T.pyriformis toxicity expressed as negative logarithm of the concentration required to inhibit 50% growth (pIGC50) T.pyriformis toxicity value >0.5 is considered toxic. Crizotinib and Apigenin both are nontoxic. Minnow toxicity is the lethal concentration values (LC50 expressed in mM) of a compound that is necessary to cause death of 50% minnows. For minnow toxicity values below 0.5 mM is considered toxic. Crizotinib predicted value is 0.942 mM, and 2.432 mM is the predicted value of Apigenin. Altogether, Apigenin is safer compound than Crizotinib (Table 4.26).

Absorption properties comparison as mentioned in the Table 4.26 shows that Water solubility of standard drug is less than the lead compound. Predicted value of water solubility of Crizotinib is less than the Apigenin but both are in safe range. Caco2 permeability predicts about the absorption of orally administered drugs. Both are in normal range. Predicted values of intestinal absorption in human are 92.006% for Crizotinib and 93.25% for Apigenin. Both compounds predict low skin permeability. Crizotinib shows 'Yes' category for P-glycoprotein substrate while Apigenin show 'Yes' category for P-glycoprotein substrate and 'No' category for P-glycoprotein I/II inhibitors model. This means Crizotinib and Apigenin as P-gp substrate show low oral absorption and p-gp I/II inhibitor reduce the pumping out of xenobiotic and toxins activity of P-gp from cell and may have high absorption.

Distribution properties are based on 4 models. The first model of distribution properties VDss (human) is the uniform distribution of the drug in blood plasma. If value higher than 2.81 L/Kg that means the drug is more distributed in tissue rather plasma. Both compounds have reasonable value of VDss.

Fu is the unbound friction in plasma. Fu value of Apigenin is more than Crizotinib that predicts it is more effective than the standard drug. BB permeability show blood brain barrier permeability, if value higher than 0.3 then drug easily cross the blood brain barrier and if the value is less than the drug may not be evenly distributed in the brain. Both the compounds have BB permeability value in tolerable range that means it provide no harm to the brain. CNS permeability j-3 is considered safe. Both compounds have CNS permeability j-3 thus considered safe.

Metabolic properties are predicted based on isoforms of cytochrome P450 which includes CYP2D6, CYP3A4, CYP1A2, CYP2C19 and CYP2C9. Crizotinib is predicted as the substrate of CYP3A4 isoform while Apigenin is not the substrate of any isoform. Crizotinib is an inhibitor of CYP1A2 and CYP3A4 isoforms but Apigenin show itself as inhibitor of only CYP1A2 isoform.

Excretion properties consist of two models with predicted values are given in Table 4.26. Drug clearance is measured by total clearance which occurs as combination of hepatic clearance and renal clearance and the value is expressed as log CL tot in ml/min/kg. Predicted value of drug clearance as total clearance of Crizotinib and Apigenin are considered safe.

Total clearance is related to bioavailability and determines the dosing rate. For Renal OCT2 substrate, Crizotinib show 'Yes' which means it interfere in the normal functioning of organic cation transporter 2 who play role in renal clearance of drugs. While Apigenin show 'No' towards Renal OCT2 substrate, this means that Apigenin does not interfere in the normal functioning of organic cation transporter.

ADMET	Model	Crizotinib	Apigenin
Properties	Name	CHZOUIIID	Apigeiiii
	AMES	No	No
	toxicity	NO	NO
	Max.	-0.095	0.328
	tolerated dose		
	(human)	mg/Kg	m mg/Kg
Toxicity	hERG I	No	No
	inhibitor	110	110
	hERG II Voc	Yes	No
	inhibitor	105	110
	Oral	3.515	2.45
	rat acute		
	toxicity	mol/Kg	mol/Kg
	Oral	1.57	2.298
	rat chronic		
	toxicity	mg/kg	mg/kg
	Hepatotoxicity	Yes	No
	Skin	No	No
	sensitization	110	110
	T.pyriformis	0.296	0.38
	toxicity	$\log\mathrm{ug/L}$	$\log\mathrm{ug/L}$
	Minnow	0.942	2.432
	toxicity	$\log \mathrm{mM}$	$\log \mathrm{mM}$
	Water	-4.14	-3.329
	solubility	$\mathrm{mol/L}$	$\mathrm{mol/L}$
	Caco2	0.702	1.007
Absorption	permeability	$\mathrm{cm/S}$	$\mathrm{cm/S}$

TABLE 4.26: ADMET properties of drug (Crizotinib) and leading compound (Apigenin)

ADMET	Model	Crizotinib	Apigenin
Properties	Name		
	Intestinal		
	absorption	92.006~%	93.25~%
	(human)		
	Skin	-2.747	-2.735
	permeability	log Kp	log Kp
	P-glycoprotein	Yes	Yes
	substrate	105	105
	P-glycoprotein	Yes	No
	I inhibitor	165	NO
	P-glycoprotein	Yes	No
	II inhibitor	105	110
	VDss	0.801	0.822
	(human)	L/Kg	L/Kg
Distribution	Fraction	0 122	0.147
	unbound	0.132 Fu	0.147 Fu
	(human)	ги	ги
	BBB	-1.164	-0.734
	permeability	log BB	$\log BB$
	CNS	-2.222	-2.061
	permeability	$\log PS$	$\log PS$
	CYP2D6	No	No
	substrate	110	110
	CYP3A4	Yes	No
Metabolism	substrate	TCD	110
	CYP1A2	Yes	Yes
	inhibitor	100	100

TABLE 4.26: ADMET properties of drug (Crizotinib) and leading compound (Apigenin)

ADMET	Model	Crizotinib	Apigenin	
Properties	Name	CHZOUIIID	Apigeiiii	
	CYP2C19	No	No	
	inhibitor		NO	
	CYP2C9	Yes	No	
	inhibitor		NO	
	CYP2D6	No	No	
	inhibitor			
	CYP3A4	Yes	No	
	inhibitor	105		
	Total	0.571	0.566	
Excretion	Clearance	$\mathrm{ml/kg}$	ml/kg	
	Renal			
	OCT2	Yes	No	
	substrate			

 TABLE 4.26: ADMET properties of drug (Crizotinib) and leading compound (Apigenin)

4.18 ADMET Properties Comparison of Paclitaxel and Salfredin B11

Model 1 of AMES toxicity depicts the standard drug and lead compound are not mutagenic. Maximum tolerated dose helps to set maximum recommended tolerated dose if value is $\leq 0.477 \log \text{mg/kg/day}$ then considered low and greater values are considered high. Table 4.27 shows 0.199 mg/kg value for Paclitaxel and -0.051mg/kg for Salfredin B11 depicts that the bio compound Salfredin B11 is safe to use. The model hERG I/II inhibitor depicts that the compounds are inhibitor of potassium channel or not. Paclitaxel show itself as hERG II inhibitor. The model Oral rat acute toxicity (LD50) expressed in mol.kg is the amount of drug that cause the death of 50% rats. LD50 value of Paclitaxel is higher than Salfredin B11. Oral rat chronic toxicity (LOEAL) determines the lowest dose of drug which can produce adverse effects of drug over long duration. LOEAL predicted value of Paclitaxel is higher than Salfredin B11 which shows its potency to be less toxic than bio compound. Hepatotoxicity indicates the injury to liver. Paclitaxel shows that it is hepatotoxic while Salfredin B11is non-hepatotoxic. Both compounds do not cause any allergic reactions. *T.pyriformis* toxicity expressed as negative logarithm of the concentration required to inhibit 50% growth (pIGC50) *T.pyriformis* toxicity value >0.5 is considered toxic. Paclitaxel and Salfredin B11both are nontoxic. Minnow toxicity is the lethal concentration values (LC50 expressed in mM) of a compound that is necessary to cause death of 50% minnows. For minnow toxicity values below 0.5 mM is considered toxic. Paclitaxel predicted value is 2.988 mM, and 1.942 mM is the predicted value of Salfredin B11. Altogether, Salfredin B11 is safer compound than Paclitaxel.

Absorption properties comparison as mentioned in the Table 4.24 shows that Water solubility of standard drug is slightly higher than the lead compound. Predicted value of water solubility of Paclitaxel is slightly higher than the Salfredin B11but both are in safe range. Caco2 permeability predicts about the absorption of orally administered drugs. Both are in normal range. Predicted values of intestinal absorption in human are 100% for Paclitaxel and 94.50% for Salfredin B11. Both compounds predict low skin permeability. Paclitaxel shows 'Yes' category for Pglycoprotein substrate while Salfredin B11show 'No' category for P-glycoprotein substrate and for P-glycoprotein I/II inhibitors model. This means Paclitaxel as P-gp substrate show low oral absorption and p-gp I/II inhibitor reduce the pumping out of xenobiotic and toxins activity of P-gp from cell and may have high absorption.

The first model of Distribution properties VDss (human) is the uniform distribution of the drug in blood plasma. If value higher than 2.81 L/Kg that means the drug is more distributed in tissue rather plasma. Both compounds have reasonable value of VDss. Fu is the unbound friction in plasma. Fu value of Salfredin B11 is more than Paclitaxel that predicts it is more effective than the standard drug. BB permeability show blood brain barrier permeability, if value higher than 0.3 then drug easily cross the blood brain barrier and if the value is less than the drug may not be evenly distributed in the brain. Both the compounds have BB permeability value in tolerable range that means it provide no harm to the brain. CNS permeability is considered safe. Both compounds have CNS permeability in normal range thus considered safe.

Metabolic properties are predicted based on isoforms of cytochrome P450 which includes CYP2D6, CYP3A4, CYP1A2, CYP2C19 and CYP2C9. Paclitaxel is predicted as the substrate of CYP3A4 isoform while Salfredin B11is not the substrate of any isoform. Paclitaxel is an inhibitor of CYP3A4 isoforms but Salfredin B11is not the inhibitor of any isoform.

Excretion properties consist of two models with predicted values are given in Table 4.27. Predicted values of drug clearance as total clearance of Salfredin B11 are high as compared to Paclitaxel. Total clearance is related to bioavailability and determines the dosing rate. For Renal OCT2 substrate, both compounds Paclitaxel and Salfredin B11 show 'No' this means that Paclitaxel and Salfredin B11 do not interfere in the normal functioning of organic cation transporter 2 who play role in renal clearance of drugs.

ADMET	Model	Paclitaxel	Salfredin
Properties	Name		B11
	AMES	No 0.199 mg/Kg	No
	toxicity		NO
	Max.		-0.051
	tolerated dose		
	(human)		m mg/Kg
Toxicity	hERG I	No	No
	inhibitor		
	hERG II	Yes	No
	inhibitor		INO

TABLE 4.27: ADMET properties of drug (Paclitaxel) and leading compound
(Salfredin B11).

ADMET	Model	Paclitaxel	Salfredin
Properties	Name	Facilitatei	B11
	Oral rat acute toxicity	2.776 mol/Kg	1.701 mol/Kg
	Oral rat chronic toxicity	3.393 mg/Kg	2.419 mg/Kg
	Hepatotoxicity	Yes	No
	Skin sensitization	No	No
	T.pyriformis	0.285	0.494
	toxicity	$\log\mathrm{ug/L}$	$\log \mathrm{ug/L}$
	Minnow	2.988	1.492
	toxicity	$\log \mathrm{mM}$	$\log \mathrm{mM}$
	Water	-3.158	-3.081
	solubility	$\mathrm{mol/L}$	$\mathrm{mol/L}$
	Caco2	0.623	1.201
Absorption	permeability	$\mathrm{cm/S}$	$\mathrm{cm/S}$
Absorption	Intestinal absorption (human)	100%	94. 508%
	Skin	-2.735	-3.236
	permeability	log Kp	log Kp
	P-glycoprotein substrate	Yes	No
	P-glycoprotein I inhibitor	Yes	No

TABLE 4.27: ADMET properties of drug (Paclitaxel) and leading compound
(Salfredin B11).

ADMET	Model	Paclitaxel	Salfredin
Properties	Name	I aciitaxei	B11
	P-glycoprotein	Yes	No
	II inhibitor	105	110
	VDss	1.458	0.363
	(human)	L/Kg	L/Kg
Distribution	Fraction		0.465
	unbound	0 Fu	0.405 Fu
	(human)		Гu
	BBB	-1.731	-0.747
	permeability	$\log BB$	$\log BB$
	CNC	-3.95	-2.827
	permeability	$\log PS$	$\log PS$
	CYP2D6	No	No
	substrate	110	110
	CYP3A4	Yes	No
Metabolism	substrate	200	1.0
	CYP1A2	Yes	No
	inhibitor		
	CYP2C19	No	No
	inhibitor		
	CYP2C9	Yes	No
	inhibitor		
	CYP2D6	No	No
	inhibitor		
	CYP3A4 inhibitor	Yes	No
	Total	-0.36	0.481
Excretion	Clearance	$\mathrm{ml/kg}$	ml/kg
	Renal OCT2 substrate	No	No

TABLE 4.27: ADMET properties of drug (Paclitaxel) and leading compound (Salfredin B11).

4.19 Physiochemical Properties Comparison

Physiochemical properties describe the basic and fundamental properties of compounds which also act as primary screeners to sort out compounds with desirable properties. Crizotinib consists of 52 atoms of carbon, hydrogen, chlorine, fluorine, nitrogen and oxygen while Apigenin consist of 30 atoms of carbon, nitrogen and oxygen which show its simplicity as a bio-compound. Molecular weight and log P value of Crizotinib is high than Apigenin. Apigenin donated more hydrogen bonds than Crizotinib. Rotatable bonds more than 10 depict decreased oral bioavailability. Crizotinib has 5 rotatable bonds as compared to Apigenin which has only 1 rotatable bond (Table 4.28).

TABLE 4.28: P	Physiochemical	properties	comparison
-----------------	----------------	------------	------------

Drug	0	Rotatable	HBD	HBA	Molecular	Molecular Wt.
	value	Bonds			Formula	(g/mol)
Apigenin	2.576	1	3	4	$\mathrm{C}_{15}\mathrm{N}_{10}\mathrm{O}_{5}$	270.24
Crizotinib	5.037	5	2	6	$\mathrm{C}_{21}\mathrm{H}_{22}\mathrm{Cl}_{12}\mathrm{FN}_5\mathrm{O}$	450.345

Paclitaxel consists of 113 atoms of carbon, hydrogen, nitrogen and oxygen while Salfredin B11consist of 29 atoms of carbon, hydrogen and oxygen which shows its simplicity as a bio-compound. Molecular weight and log P value of Paclitaxel is high than Salfredin B11. Paclitaxel and Salfredin B11 donated equal hydrogen bonds. Rotatable bond more than 10 depicts decreased oral bioavailability(Table 4.29).

TABLE 4.29: Physiochemical properties comparison.

Drug	0	Rotatable Bonds	HBD	HBA	Molecular Formula	Molecular Weight (g/mol)
Salfredin	2.2468	0	1	4	$C_{13}H_{12}O_4$	232.32
B11	2.2400	0	T	4	$0_{13}11_{12}0_{4}$	232.32

Drug		Rotatable Bonds	HBD	HBA	Molecular Formula	Molecular Weight (g/mol)
Paclitaxel	3.7357	10	14	4	$\mathrm{C}_{47}\mathrm{H}_{51}\mathrm{NO}_{14}$	853.9

TABLE 4.29: Physiochemical properties comparison.

4.20 Docking Score Comparison

Discovering protein-ligand binding site and conformations are important in drug discovery. Therefore standard drug as ligand was docked against the selected receptors by CB-Dock online tool which predicts the cavities of protein and calculates the centers and sizes of the top 5 cavities for all the three proteins. Results of docking of standard drugs and lead compound selected against the two receptors namely ALK and EML4 receptors are shown in tables. The highest binding score shown by Apigenin is -7.9 against ALK which is less than Crizotinib that shows -8.5 against the same protein ALK. The highest binding score shown by Salfredin B11 is -7.2 against ALK which is less than Paclitaxel that shows -8.2 against the same protein EML4. The interaction visualization analysis studies are performed by PyMol molecule visualization tool and Ligplot+ (V.1.4.5).

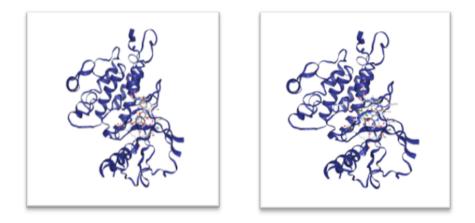


FIGURE 4.17: Best Pose Interaction of Apigenin and Crizotinib as Ligand with ALK.

4.21 Docking Analysis Comparison

4.21.1 Docking Analysis of Drug and Lead Compound with ALK

Best docking score of reference drug and lead compound are analyzed by LIG-PLOT+ (V.1.4.5), (Figures 4.19 and 4.20). Docking results are analyzed on the basis of;

- 1. No. of Hydrogen bonds.
- 2. No. of steric interactions.
- 3. No. of interacting amino acids.
- 4. Interaction with hydrophobic regions.

The detail of hydrogen bonds and hydrophobic interactions are displayed on Table 4.30. Oxygen atoms present in ligand play important role in H-bond formation with target protein. Apigenin makes 2 hydrogen bonds whereas Crizotinib makes 3 hydrogen bonds. Furthermore, hydrophobic interactions are more in number in Crizotinib as compared to Apigenin.

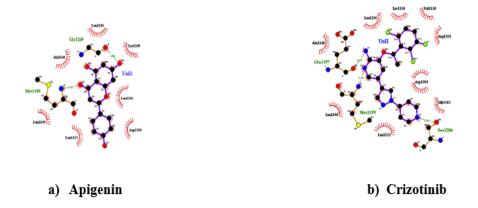


FIGURE 4.18: H- bonds and Interaction of Apigenin and Crizotinib as Ligand with ALK.

	Binding	No.	Amino		Hydrophobic
Ligands	Energy	of H.B	Acids	Distance	Interaction
					Leu1196
					Lys1150
					Ala1148
			Mot1100	2.06	Leu1198
Apigenin	-7.9	2	Met1199	2.96	Leu1122
			Gly1296	2.26	
					Asp1203
					Leu1256
					Ala1148
					Leu1256
					Leu1198
			Met1199	2.93	Lys1150
Crizotinib	-8.7	3	Ser1206	3.00	Val1130
			Glu11897	3.05	Leu1122
					Gly1202
					Arg1253
					Asp1203

TABLE 4.30: Hydrogen Bonds and Interactions comparison of Apigenin and Crizotinib.

4.21.2 Docking Analysis Comparison Docking Analysis of Drug and Lead Compound with EML4

The 2D diagrams generated through Ligplot by the interaction of Paclitaxel and Salfredin B11 with EML4 is shown in Figure. And the properties are mentioned in Table 4.31.

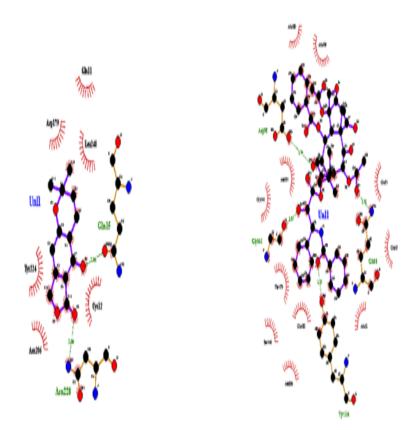


FIGURE 4.19: H- bonds and Interaction of Salfredin B11 and Paclitaxel as Ligand with EML4

The detail of hydrogen bonds and hydrophobic interactions are displayed on Table 4.31. Oxygen atoms present in ligand play important role in H-bond formation with target protein. Salfredin B11 makes 1 hydrogen bond whereas Paclitaxel makes 4 hydrogen bonds. Furthermore, hydrophobic interactions are more in number in Paclitaxel as compared to Salfredin B11.

Ligands	Binding Energy	No. of H.B	Amino Acids	Distance	Hydrophobic Interaction
					Ala1148
					Leu 1256
Salfredin					Leu1198
B11	-7.4	1	Met1199	2.88	Lys1150
DII					Val1130
					Leu1122
					Gly1202
					Gln15
					GluT1
					Ala12
	-7.3		Gly143	3.01	Ser140
Paclitaxel		4	Asp98	3.90	Glu71
			Gln11	1.92	Asp98
			Tyr334	3.90	Ala180
					Thr 179
					Asn206
					Gly142

TABLE 4.31: Hydrogen Bonds and Interactions comparison of Salfredin B11 and Paclitaxel.

Chapter 5

Conclusions and Recommendations

The motive of the present study was to discover the active constituents from the medicinal plants Nigella sativa which could act as anticancer agents in Lungs Cancer. For this purpose, 15 ligands were selected after performing studies on literature databases and docked against receptor proteins which are ALK and EML4, involved in lung cancer. The structures of all the 15 ligands were easily available in PubChem and proteins structures were also available in Protein Data Bank. Drug likeliness of compounds was studied and reported by using primary and secondary filter (Lipinski's rule of 5 as primary and pharmacokinetics properties as secondary filter). The docking procedures were performed using CB-Dock online tool. The results were visualized using PyMol and were analyzed through Ligplot version v.1.4.5. After detail analysis of their binding score, physiochemical properties and ADMET properties, Apigenin were selected as lead compound against ALK, Salfredin B11 were selected as lead compound against EML4. Virtual screening results, physiochemical properties and pharmacokinetics properties of these compounds were compared with an FDA approved drugs namely Crizotinib and Paclitaxel. Based on results, it was found that selected lead compounds show better binding affinity to respective protein targets and show less toxicity than standard drugs.

5.1 Recommendations

Lead compounds Apigenin and Salfredin B11 as per this research results should be explored as a drug candidate for the treatment of Lungs cancer.

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